

A Leach of Faith: Moving ASGM Beyond Cyanide

Alternative lixiviants in artisanal and small-scale gold mining: the evidence on recovery, cost, exposure and scale, and what must be in place whatever the reagent

Lixiviant: the leaching solution that dissolves gold out of crushed ore, concentrate or tailings and carries it into solution, where it can then be recovered.



The test: A 'cyanide-free' label taken on trust is a leap of faith. The test is the same for every lixiviant: regulated operators, sound chemicals management to protect human health and the environment. Evidence, not claims.

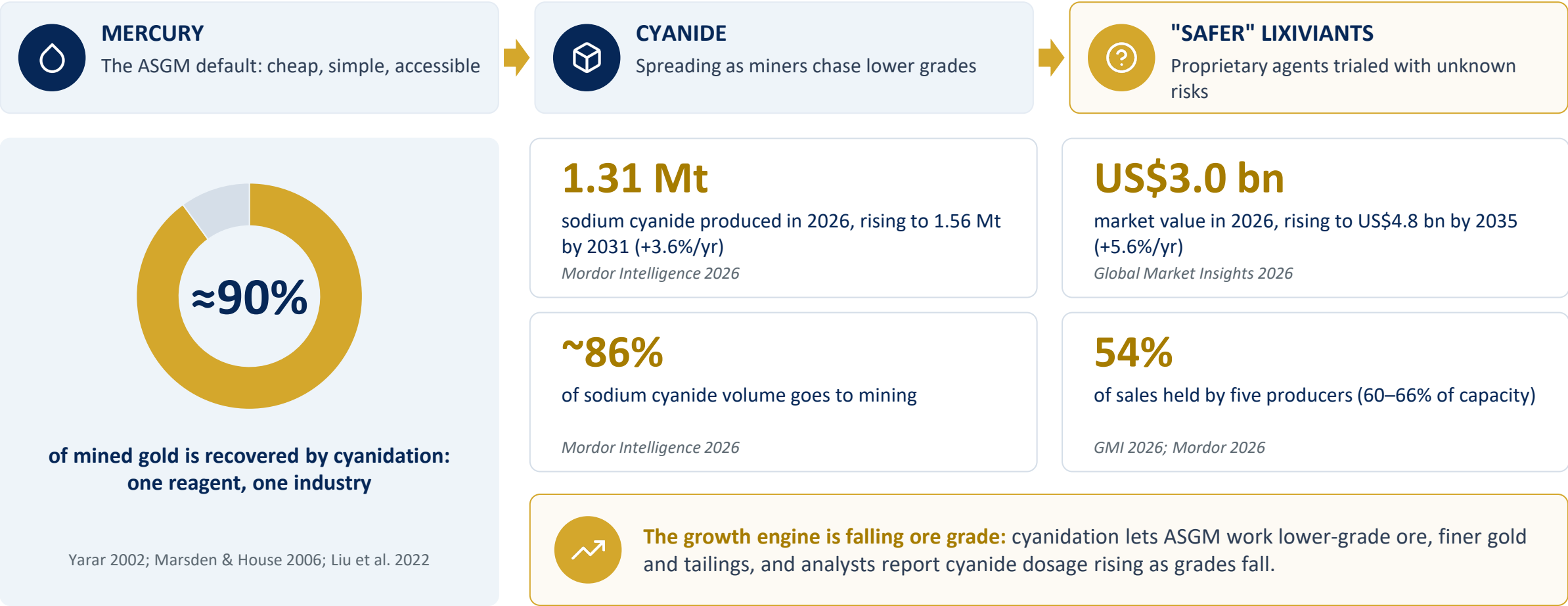
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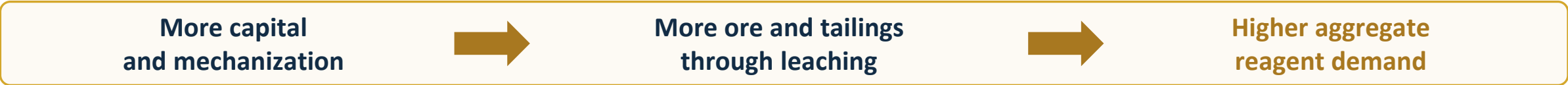
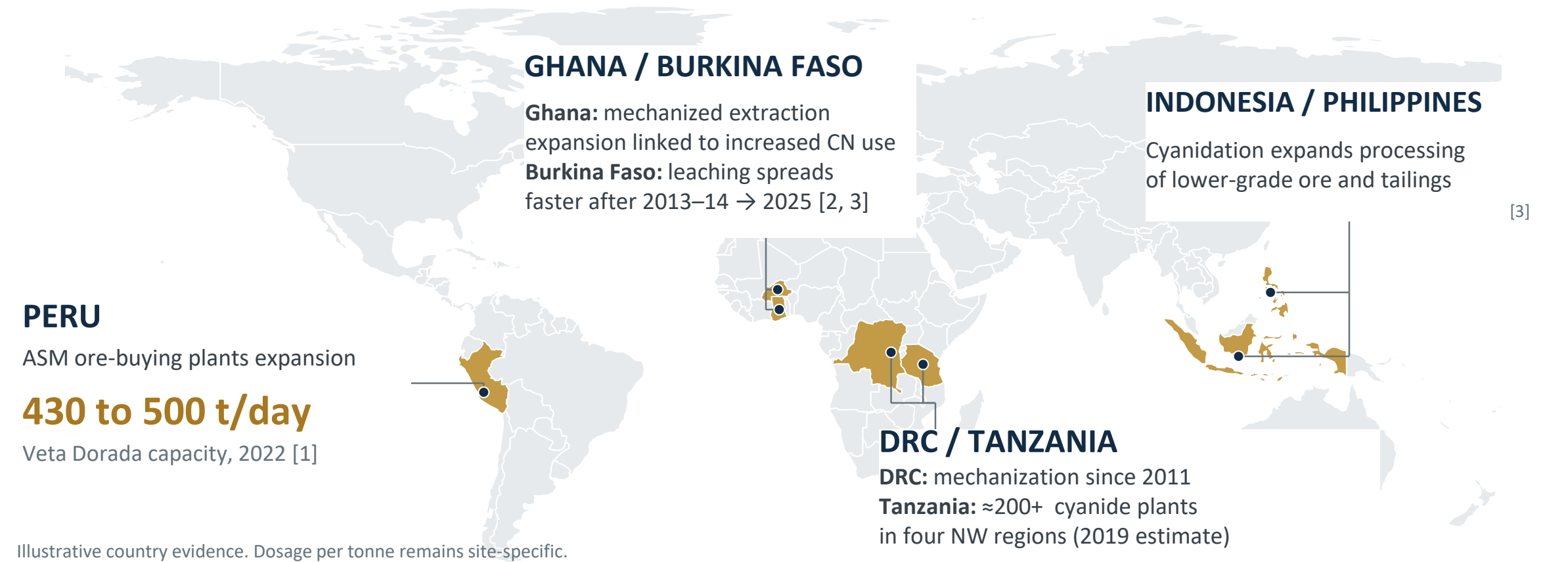
Cyanide is the world's gold lixiviant, and it is growing

Mercury is the ASGM default; cyanide is spreading as grades fall; "safer" proprietary agents are arriving unverified



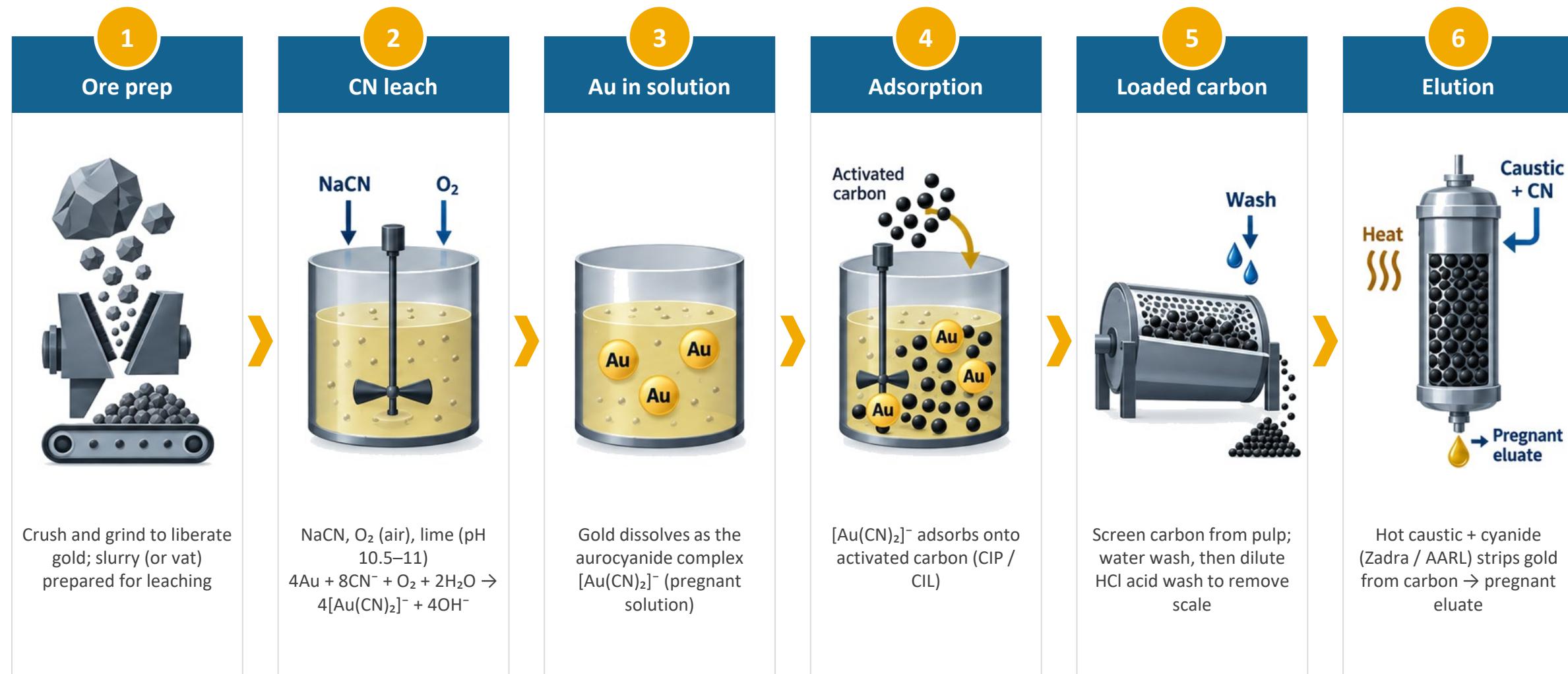
Semi-industrial mining growth is expanding gold leaching

Documented shifts in mining scale and processing across seven countries: more capital, more tonnes through leaching, more reagent



How cyanide leaching works: the circuit every alternative must fit

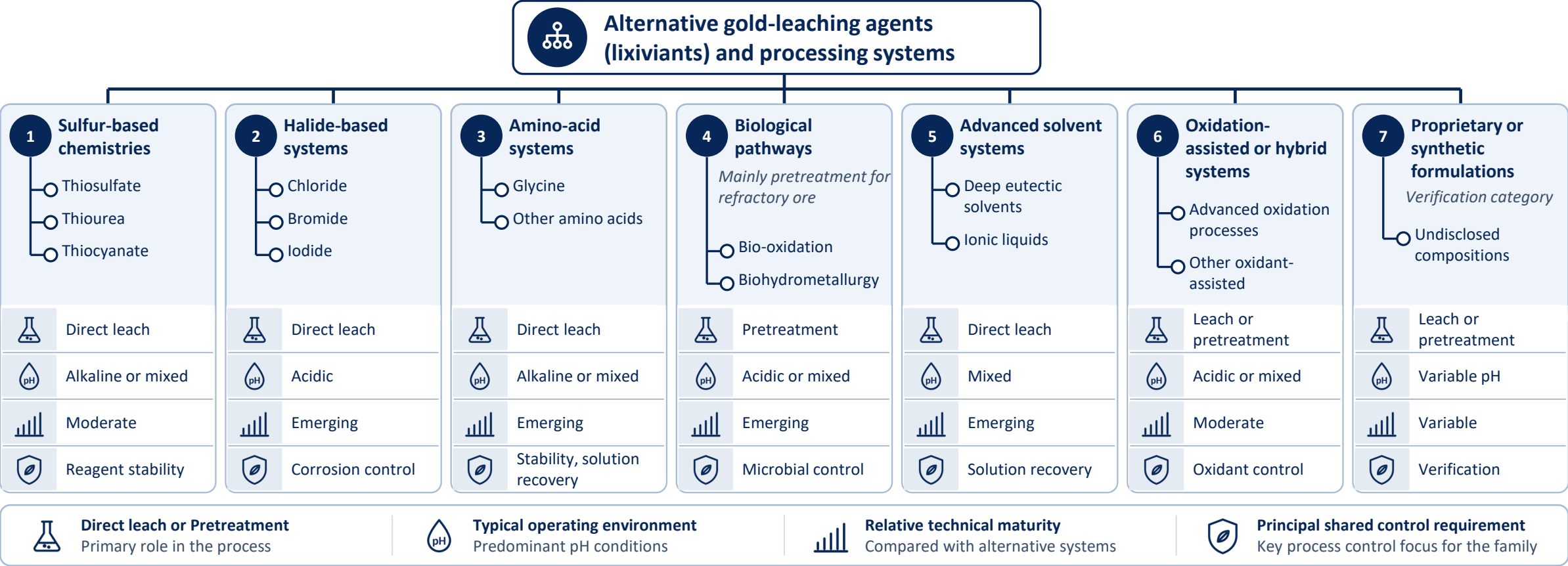
From ore to pregnant eluate: simplified CIP / CIL schematic. Jinchan, GlyCat and cassava all run this same alkaline circuit.



Source note: Chemistry per Elsner's equation; protective alkalinity (lime, pH 10.5–11) prevents HCN evolution; elution by Zadra or AARL. Not shown: electrowinning or zinc precipitation, smelting, carbon regeneration. In ASGM, vat (percolation) leaching commonly replaces agitated tanks; the chemistry is the same (Marsden & House 2006; Stapper, Dales, Velasquez & Keane 2021).

Seven families of alternative gold-leaching systems

They share chemical characteristics but differ in maturity, process control, waste management and ASGM suitability



Fourteen questions, asked of every option

A due-diligence screen answered with cited evidence, not a numerical score (Report 1 §5 and §6)





Same questions for every option, every rating traceable. The result ranks readiness for a central hub is a due diligence screen – not a scoring model.

High recovery does not mean ASGM-ready

Readiness by technology family, graded and evidence register; color is a screening grade, never a safety rating

Technology family	Technical promise	Maturity	Process complexity	Delivered cost	Control burden	ASGM suitability	Positioning
Cyanide (baseline)	Strong	Mature	Moderate	Strong (low)	Low (high concern)	Only if governed	Benchmark: govern, do not assume
Thiosulfate	Strong	Moderate	Limited (high)	Moderate	Limited (high)	Moderate	Most credible near-term alternative, in controlled hubs only
Glycine and amino acids †	Moderate	Emerging–moderate	Moderate	Moderate	Moderate	Moderate	Lower-risk pilot pathway in contained hubs
Bio-oxidation	Moderate	Moderate	Limited (high)	Moderate	Limited (high)	Low to moderate	Pretreatment for refractory ore; central facilities
Thiourea	Moderate–strong	Limited	Limited (high)	Low (high cost)	Low (high concern)	Low	Technically interesting, weak ASGM fit
Thiocyanate	Moderate	Emerging	Limited (high)	Low (high cost)	Limited (high)	Low	Promising chemistry, not yet mature
Halides	Strong	Limited	Limited (high)	Low (high cost)	Low (high concern)	Low	Highly engineered central facilities only
Advanced oxidation	Emerging	Emerging	Limited (high)	Insufficient evidence	Limited (high)	Low	Future R&D, not deployable
DES and ionic liquids	Emerging–moderate	Emerging	Limited (high)	Low (high cost)	Limited (high)	Low	Emerging research only
Proprietary agents (GDAs) *	Moderate	Moderate (mixed)	Moderate	Moderate	Limited (high)	Low to moderate	Held at the gate: verify composition before any use

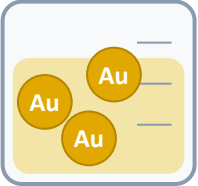
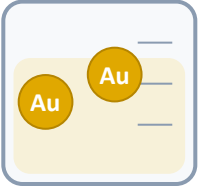
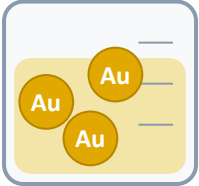
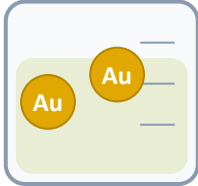


Laboratory recovery is not field performance. The most credible pathway is controlled, centralized processing, not miner- or village-level chemical use.

* Cyanide-bearing on independent assay (Deng et al. 2026). † GlyCat (glycine + cyanide) is rated within this family; its control requirements are those of cyanide at reduced concentration.

Four options worth piloting, each with one condition

Selected on independent evidence and ASGM applicability; the rest are research chemistries or engineered-plant chemistries

 Thiosulfate $S_2O_3^{2-} \cdot Cu-NH_3$ or Ca system  Benefit Beats cyanide where cyanide fails: 90.5% vs 61.7% on carbonaceous, preg-robbing ore (Ethiopia, laboratory). Limitation Reagent bill 2–6× cyanide; gold needs resin recovery; ammonia effluent. Ran at Goldstrike 2014–23, then reverted to cyanide. Maturity One commercial plant, 2014–23; none today Instrumented, supervised hub only	 Alkaline glycine $NH_2CH_2COO^-$ + oxidant · cyanide-free  Benefit No cyanide at any point; low toxicity; the carbon columns already on site recover the gold. Limitation Slow: about half of cyanide's recovery on real ASGM tailings (45% vs 87%, Cameroon, laboratory). Needs an oxidant. Maturity Pilot and demonstration scale (TRL 6–8) Contained hub pilots on real ASGM feed	 GlyCat Glycine + cyanide at 20–35% of dose  Benefit 65–80% less cyanide at comparable recovery: Bulyanhulu CIL and Bayan Airag heap (operator-reported). Limitation Still cyanide: custody, WAD monitoring and destruction apply. Large-mine data, unaudited; no hub trial yet. Maturity Commercial at large mines since 2023–24 Where cyanide is already used	 Bio-oxidation Bacterial pretreatment of sulfide ore  Benefit Unlocks refractory sulfide ore before any leach; mature at industrial scale (TRL 8–9). Limitation A pretreatment, not a lixiviant: slow, acidic, needs microbial control; the leach that follows still needs a reagent. Maturity Commercial at large mines; central plants Central-plant pretreatment only
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Maturity scale: laboratory · pilot · commercial at large mines · proven on ASGM feed (green). None of the four reaches the last segment.



Faster leaching ≠ higher recovery ≠ safer process. None is proven on ASGM feed; each is a hub technology with its own control requirement. Ore-specific test work first.

Is Jinchan cyanide in disguise?

What the label says, against what independent laboratories find



WHAT THE LABEL SAYS



"Cyanide-free", "eco-friendly", "non-toxic"



Drop-in substitute for sodium cyanide: same circuit, same carbon



"Ordinary cargo": no dangerous-goods rules for transport or storage



High recovery, on the supplier's own test data



WHAT INDEPENDENT LABORATORIES FIND



A cyanide circuit: pH 10–12, carbon or zinc recovery (Zhang et al. 2022; Deng et al. 2026)



Free cyanide in the reagent before it touches ore: 7,300 mg/L CN^- in a 100 g/L solution; 41 mg/L still in the leachate (Deng et al. 2026)



Slower than cyanide: 91.7% vs 97% at 24 h in Ghana; 48 h to catch up, at 1.6× the dose (Beyuo et al. 2016)



Hungrier: 1.3–2.3× more reagent per ton of ore in the Ghana and Turkish tests (Beyuo et al. 2016; Yilmaz & Sayin 2024)



The most dangerous property for ASGM: more reagent can mean less gold. Above about 400 ppm, adding Jinchan reduced gold yield (Beyuo et al. 2016). Where dosing is by eye, "add more" makes it worse and loads cyanate and ferrocyanide into the tailings (Deng et al. 2026).



ANSWER: presume cyanide-bearing or cyanide-generating until an independent assay says otherwise. A confidential composition means no risk assessment is possible; "ordinary cargo, no restrictions" is an evasion flag.



Source note: Deng et al. 2026 (Hydrometallurgy 244:106799; independent laboratory speciation, Curtin); Beyuo et al. 2016 (Gold Fields Ghana, laboratory); Yilmaz & Sayin 2024 (Turkish ores, laboratory); Zhang et al. 2022 (review). Supplier claims are marketing material, not performance evidence.

Five samples settle it

Before any program touches a proprietary agent: sample the circuit, speciate at an independent laboratory



WHAT THE LABORATORY MEASURES IN EVERY SAMPLE

Free cyanide (CN⁻) The acutely toxic form: HCN gas below pH 9.	WAD cyanide Weak-acid dissociable: free cyanide plus weakly bound copper, zinc and nickel complexes, regulatory measure.	Total cyanide Every form, including stable iron complexes, which release free cyanide in sunlight within minutes.	Also Cyanate, thiocyanate, ammonia and metals: the by-products that show what the reagent really is.
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 **Rules:** an independent, accredited laboratory, not a colorimetric field kit · preserve samples (pH above 12, cold, dark) · report the method with every result

 **Cyanide found: govern the reagent as cyanide.**
Not found: the claim has been tested; the reagent is still not endorsed until recovery, effluent and cost are shown on real ASGM feed.

The product's own documents say "cyanide-class"

Patent, stock-exchange filings and safety data sheet, read together with the independent chemistry



THE PATENT ON THE PACK

- **WO2012079283A1**: ferrocyanide fused with urea and alkali at 600–1,000 °C.
- **Stated LD₅₀ 50 mg/kg** (mice): GHS acute oral Category 3, toxic.
- **Every patent in the class** uses the same cyanide-salt roast (Zhang et al. 2022, Table 6).



THE MANUFACTURER'S FILINGS

- **Product class**: "environment-friendly cyanide-class reagent"; active ingredient contains cyano groups (Senhe listing documents, 2024–26).
- **No national or industry standard** for the class; not a hazardous chemical in China.
- **Exported as "chemical preparations"** (HS 3824.99), not "cyanides" (HS 2837): invisible to cyanide import controls.



THE CHEMISTRY IN SERVICE

- **At the recommended dose** (0.3–0.8 g/L) the reagent gives about 20–60 mg/L free cyanide before any ore contact: a low-strength cyanide circuit, near the 50 mg/L WAD limit for open ponds.
- **"Chelated, cannot ionize" fails**: free cyanide was measured in the beaker (Deng et al. 2026), and iron–cyanide complexes release it in sunlight within minutes (Johnson et al. 2002).
- **The safety data sheet itself** prescribes hypochlorite and ferrous sulfate for the wastewater: cyanide destruction chemistry.



IMPLICATION: treat Jinchan, CNFREE, CNLITE and generic gold dressing agents as cyanide-class reagents for handling, storage, effluent, worker training and Minamata Annex C. Their marketed advantage is regulatory, not metallurgical.



Source note: planetGOLD technical note (Sept 2026), Jinchan global evidence review. Patent WO2012079283A1 (number printed on the packaging); Guangxi Senhe BSE listing documents (2024–26, read in translation); Jinchan SDS TBR-C-202402-0120-1; Deng et al. 2026; Johnson et al. 2002; Zhang et al. 2022. Where the class is sold and used: slide 15 and backup B4.

Glycine cuts cyanide; it does not yet replace it

Two products are sold as "glycine": one is cyanide-free and slow, the other contains cyanide by design



ALKALINE GLYCINE: cyanide-free

- Glycine, alkali and an oxidant. No cyanide at any point.
- On real ASGM tailings: ~45% recovery in 10 h vs 87% for cyanide in 24 h on the same material (Cameroon; Mohamed et al. 2025, laboratory).
- No acutely lethal reagent to store or destroy; the oxidant is the remaining hazard, and it also consumes the glycine.



GLYCAT: glycine + cyanide, by design

- Laboratory: 0.5–1 M glycine with 500 ppm cyanide gave 90% gold where 500 ppm cyanide alone gave 46%: about 80% less cyanide for the same gold (Barani et al. 2021).
- At scale (operator-reported, unaudited): Bulyanhulu CIL, 80% cyanide cut at comparable recovery; Bayan Airag trial heap, 65% cut, 78% recovery.

WHAT IT MEANS FOR A SMALL-SCALE HUB



On-site carbon columns recover the gold: 13.2 g Au/kg carbon in 4 h (Oraby & Eksteen 2015, laboratory)



Not dangerous goods; a permitted food additive (FDA 21 CFR 172.812; EU E640)



Cheaper per ton of reagent: about US\$1,700 vs US\$2,380 for cyanide (2025–26 indices)



About half the recovery of cyanide on real ASGM tailings (Mohamed et al. 2025)



Needs an oxidant, which is the remaining hazard and degrades the glycine as it works (Oraby et al. 2020)



Consumption on ASGM feed not established; recyclability unvalidated (Fashu et al. 2026)

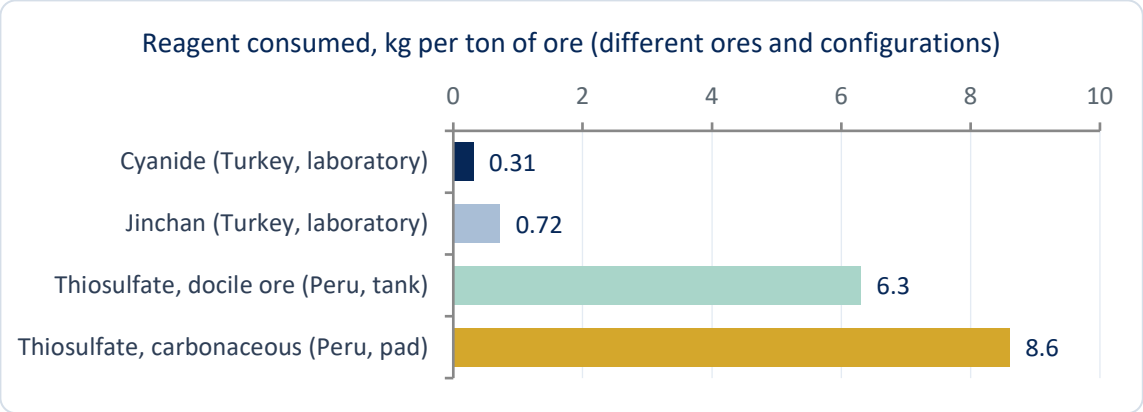
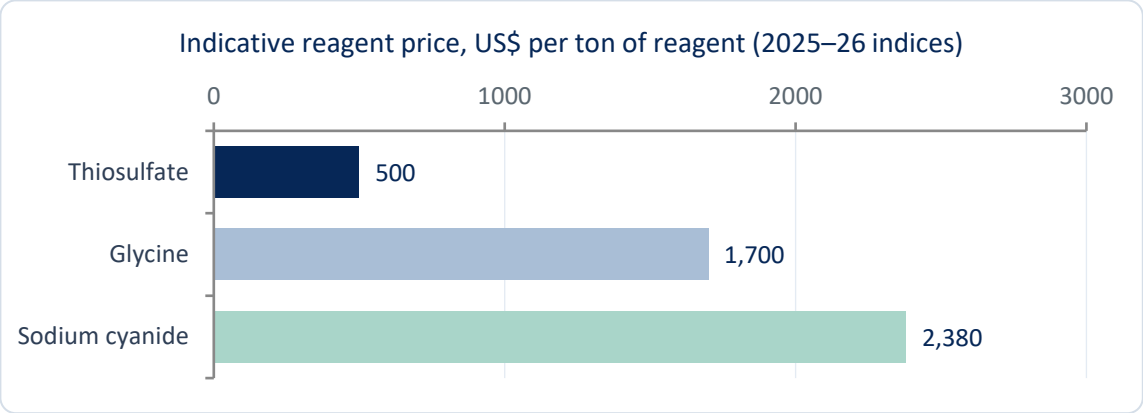


ANSWER: yes, at large-mine scale GlyCat cuts cyanide 65–80% at comparable recovery (pending independent audit). It is cyanide reduction, not replacement: custody, control and destruction still apply. Stand-alone glycine is the cyanide-free option, and it is slow.



Thiosulfate: mature chemistry, instrumented plants only

The most mature non-cyanide lixiviant and the right tool for carbonaceous ore; not a yard reagent



THE VERDICT, LINE BY LINE



- Beats cyanide where cyanide fails: 90.5–91.5% vs 61.7% on Ethiopian carbonaceous ore (Chaka & Rupprecht 2024, laboratory).
- The only non-cyanide lixiviant run at full industrial scale on whole ore: Barrick Goldstrike, 2014–2023.
- Goldstrike converted back to cyanide in 2023 (US\$30 M) for higher recovery and lower cost; no commercial plant runs it today (Nevada Gold Mines 2025).
- Consumes 6.3–8.6 kg/t: about US\$3–8 per ton of ore against US\$0.7–3.6 for cyanide (Tremolada 2018; 2025–26 indices).
- Gold will not load on the carbon ASGM centers own: needs resin or electrowinning; ammonia and high-COD effluent with no standard destruction circuit (Aylmore 2016).



ANSWER: of age for carbonaceous or preg-robbing ore, in an instrumented, supervised plant with resin recovery and effluent treatment. Not for a yard, and no longer proven at any operating plant.



Source note: Prices: IMARC / ChemAnalyst 2025–26 indices (thiosulfate US\$475–890/t; NaCN US\$1,700–3,300/t by region), indicative. Consumption bars compare different ores, scales and configurations (Yilmaz & Sayin 2024, laboratory; Tremolada 2018, Peru tank and pad): indicative, not like-for-like. Ecuador thiosulfate–glycine: Redrovan et al. 2025.

Every alternative changes the hazard; yet vulnerable populations stay the same

"Cyanide-free" is a label, not a safety finding: women and children have the highest solution contact and the least protection

Reagent	Acute hazard	Effluent and residue	Manageable in a hub?
Cyanide	HCN gas; ingestion; acutely lethal	Proven destruction circuits (SO ₂ /air, peroxide, hypochlorite)	Yes, only if governed
Thiosulfate	Low acute toxicity; ammonia in the copper–ammonia system	No standard circuit; heavy COD load	Heavy burden
Alkaline glycine	Low acute toxicity; food-grade reagent	Biodegradable; residues untested on mercury tailings	Lightest of any option
GlyCat	Cyanide present at 20–35% of normal dose	Cyanide destruction still required	Manageable in a hub
Jinchan / GDAs	Undisclosed; cyanide-bearing or -generating	Cyanate and ferrocyanide in leachate	Cannot be assessed
Thiourea · halides · polysulfides	Suspected carcinogen (thiourea); chlorine gas; H ₂ S on pH excursion	Acid or halogen effluent; gas control	No

**WHO HANDLES THE PULP**

**Women:** concentrated in ore dressing, panning and processing: the highest solution contact, the least protective equipment (Hinton, Veiga & Beinhoff 2003; Dales personal observation 2024-26)

**Children:** an identified exposed group at and around processing sites (Kohio et al. 2025; Stapper et al, 2021).

**Communities:** water and soil beside processing yards: Burkina Faso, 9% of ASGM-area water samples over the WHO free-cyanide guideline, peak 2.42 mg/L (Kohio et al. 2025)



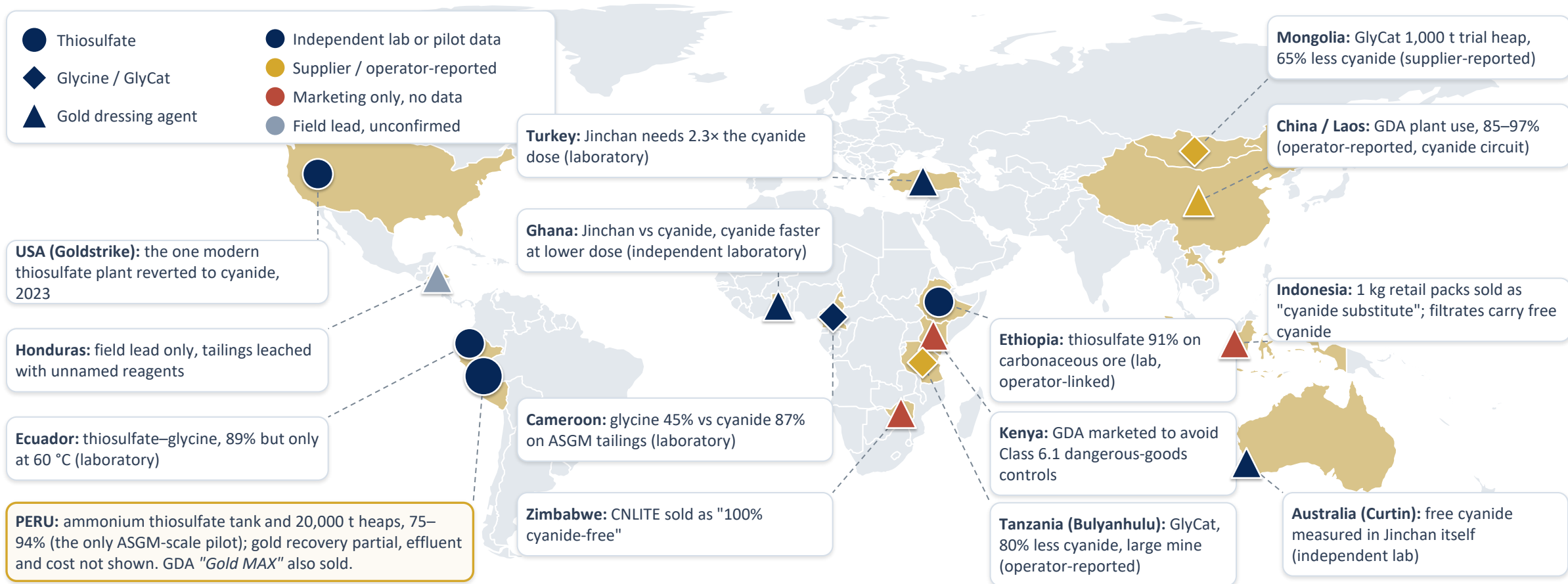
Kenya (Kakamega–Vihiga): cyanidation plants operate beside mercury processing; soil and sediment mercury, chromium and nickel exceed guidelines; cyanide itself has not been measured (Ondayo et al. 2023). Every alternative swaps one waste profile for another, often a less regulated one.



A reagent evaluation that ignores who physically handles pulp is incomplete. PPE, training and rotation for pulp handlers apply to every reagent, branded or not.

Every case dissolves gold; none crosses the wall

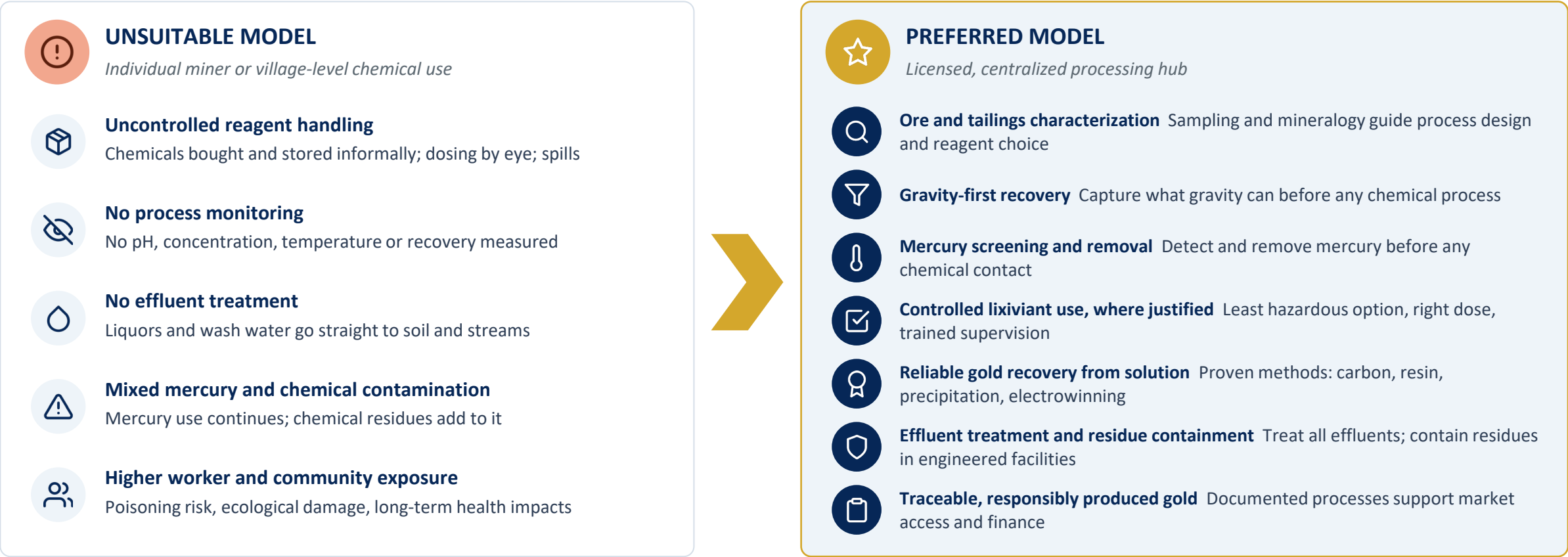
Principal finding or risk in each place; marker shape is the technology, marker color is the strength of the evidence



THE WALL: no ASGM-relevant case has shown gold recovery from solution, effluent management and delivered cost on ASGM feed. Only the large-mine GlyCat cases cross it, on supplier data.

From chemical substitution to controlled processing systems

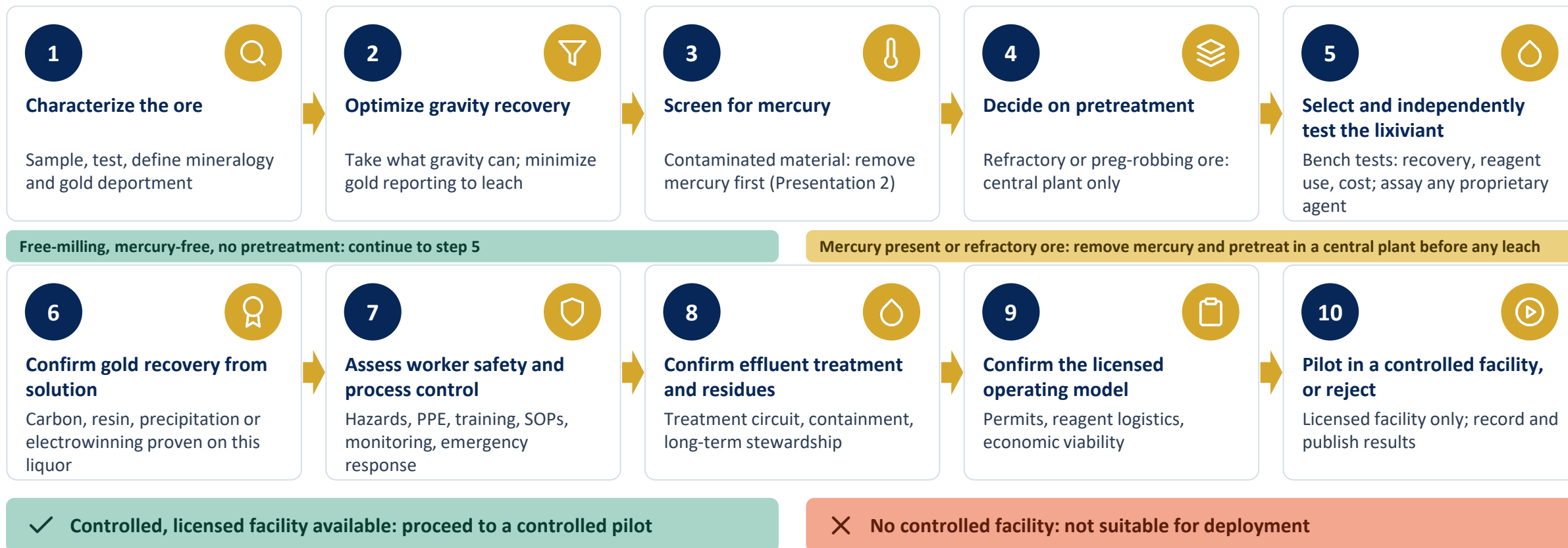
Safety and stewardship come from the system, not from the reagent



 **The transition is not to replace cyanide;** it is to move chemical processing into controlled systems. Centralized, supervised plants protect people, communities and the environment.

From ore characterization to a processing decision

Ten steps, none optional and none reagent-specific: a "cyanide-free" label skips none of them



★ **Lower toxicity does not mean inherently safe:** process control, residue management and governance remain essential, whatever the reagent.



Source note: Infographic 4 of the planetGOLD alternative-lixivants series; framework synthesized from the Report 1 literature review and planetGOLD program framing. Mercury removal before any leach: Minamata Convention Annex C 1(b)(iv); Veiga et al. 2020.

What the evidence means, and where we go from here

Three conclusions for decision-makers, and the first moves for each audience

1

No alternative lixiviant is ASGM-ready across the range of ores, settings and operating realities.

2

The credible pathway is controlled, centralized processing with strong governance and environmental safeguards.

3

Decisions must rest on independent evidence and be context-specific, not driven by marketing claims.



OPERATORS

- Gravity first: pre-concentrate before any leach and cut the mass needing chemistry.
- A contained, alkaline circuit with the recovery train (carbon or resin) specified before the first ton.
- Reagent custody: licensed supply, locked storage, dosing by titration; PPE, training and rotation for whoever handles the pulp.



REGULATORS

- License the chemistry, not the brand: Jinchan, CNFREE, GlyCat and cassava fall under cyanide rules; watch imports under HS 3824.99 as well as HS 2837.
- Set the numbers: free 0.1, WAD 0.5, total 1.0 mg/L in effluent, WAD below 50 mg/L in any open pond, and a laboratory that can measure them.
- Permit and inspect centralized plants; apply Minamata Annex C to cyanide-generating agents on mercury-bearing tailings.



SUPPORT PROGRAMS

- Assay before you endorse: the five-sample independent speciation of any proprietary agent.
- Invest in centralized, supervised plants and borrow the Cyanide Code's verification and corrective-action loops in a subsidized form.
- Demonstrate the whole flowsheet, leach + recovery + effluent + delivered cost, on real ASGM feed before scaling anything.



Sequence: responsible cyanidation first; alternatives only where leach, recovery, effluent and delivered cost have been demonstrated on real ASGM feed. Replacing the reagent does not replace the management system.

Backup slides

Supporting detail, reference only: not part of the 20-minute talk

Reading	Suggested reading, prioritized	B8	Traceability behind the ratings (2 of 2)
B1	Amino-acid chemistry: why glycine	B9	The key finding in one line, by family
B2	Evidence register: scale and independence behind every number	B10	Where the evidence stops: seven countries, one wall (table)
B3	Independent assay protocol: the five samples	B11	What must be in place: controls by audience
B4	Jinchan and the gold dressing agents by country	B12	Recommendations and priority evidence gaps
B5	Full assessment matrix: 14 criteria	B13	Where the options fit: transition ladder
B6	The fourteen criteria and the four-tier classification	B14–B17	References with DOIs and stable links
B7	Traceability behind the ratings (1 of 2)		

Suggested reading, prioritized

Attached papers and the key reviews first, then the primary studies cited on the slides

Reference	Why read it	Link
Deng, Schwehr, Yang & Oraby (2026) A comparative study of cyanide, Jinchan and GlyCat™ for gold ore leaching. Hydrometallurgy 244:106799.	The independent speciation evidence behind Question 1 and the GlyCat comparison (laboratory).	doi.org/10.1016/j.hydromet.2026.106799
Aylmore (2016) Alternative lixiviants to cyanide for leaching gold ores. In Gold Ore Processing, 2nd ed., pp. 447–484.	The reference review of all alternative chemistries, conditions, consumption and cost.	doi.org/10.1016/B978-0-444-63658-4.00027-X
Zhang, Cui, Wang, Liu & Lyu (2022) A review of gold extraction using alternatives to cyanide: the novel eco-friendly synthetic gold lixiviants. Minerals Engineering 176:107336.	The key review of the Chinese synthetic reagent class: sixteen lixiviant classes, the patent recipes (ferrocyanide + urea + alkali), operator-reported plant results, and the admission that composition and mechanism remain unclear.	doi.org/10.1016/j.mineng.2021.107336
Stapper, Dales, Velasquez & Keane (2021) Best Management Practices for Cyanide Use in the Small-Scale Gold Mining Sector. planetGOLD.	The control framework behind Objective 2: the test every lixiviant must pass.	planetgold.org
Surimbayev, Bolotova, Akcil & Ghorbani (2026) Gold mining tailings as a sustainable secondary resource. Mineral Processing and Extractive Metallurgy Review.	Technology-readiness ratings for every alternative; why non-cyanide routes still need industrial validation.	doi.org/10.1080/08827508.2026.2693551
Li, Sun, Kou, Wang & Liu (2025) Development of a gold leaching reagent as an alternative to cyanide. Int. J. Miner. Metall. Mater. 32(4):835–850.	What the synthetic "eco-friendly" reagent class is made of, and why its composition is contested.	doi.org/10.1007/s12613-024-2957-x
Chaka & Rupprecht (2024) Thiosulfate leaching in carbonaceous gold-bearing ores in Ethiopia. Scientific Reports 14.	The African carbonaceous-ore case for thiosulfate (laboratory).	doi.org/10.1038/s41598-024-69646-3
Mohamed, Oupoungam, Boyraz, Alp & Sinon (2025) Glycine and cyanide leaching of artisanal gold mining tailings. Physicochem. Probl. Miner. Process. 61(3).	The only direct glycine-vs-cyanide test on ASGM tailings (Cameroon; unequal leach times).	doi.org/10.37190/ppmp/205177
Kohio, Sossou, Karoui & Yacouba (2025) Environmental cyanide pollution from artisanal gold mining in Burkina Faso. IJERPH 22(7):1125.	Measured community exposure to cyanide around ASGM processing.	doi.org/10.3390/ijerph22071125
Reno, Aulia & Dian (2026), supplier-affiliated Sodium cyanide replacement with polymerized cyanide GOLDIX. E3S Web of Conferences 693:02007.	An example of how the cyanurate class is presented; note the free cyanide it measures.	doi.org/10.1051/e3sconf/202669302007
Teng, Huang, Long & Fang (2010/2012), patent Mineral-dressing agent for noble metal. WO2012079283A1 / CN102321807B (the patent number printed on Jinchan packaging).	The ferrocyanide–urea formulation and the LD ₅₀ of 50.1 mg/kg, disclosed in the patent on the packaging.	patents.google.com/patent/WO2012079283A1

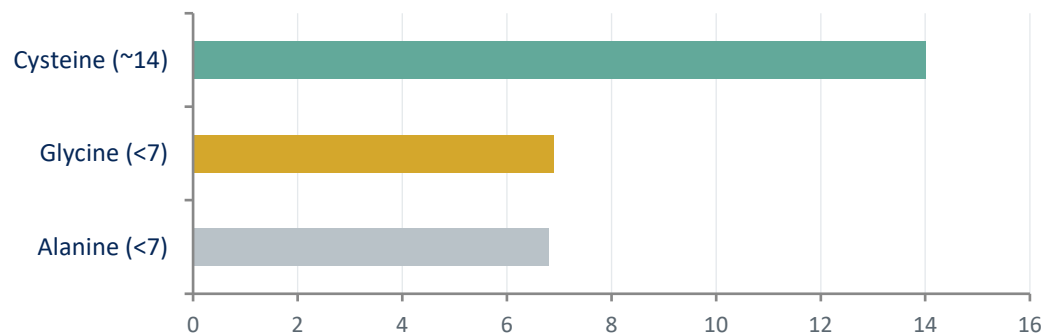


Source note: Full reference list with DOIs: backup slides and speaker notes. Every figure on the slides traces to a row of the evidence register (backup B2) or the full assessment matrix (backup B5).

Glycine is one amino acid among many, and not the strongest

Two different things are compared in marketing and in the literature: how fast an amino acid dissolves gold, and whether it survives a circuit

Gold dissolved from pure gold, matched conditions (mg/L): 1983 data, indicative



Cysteine dissolves gold about twice as fast as glycine or alanine (cysteine ≈14 mg/L; glycine and alanine <7 mg/L). Korobushkina et al. 1983, as summarized by Fashu et al. 2026; bar lengths indicative.



So why does industry use glycine?



Every other amino acid polymerizes or degrades in service; cysteine is faster but unstable, and costly (Sarvar et al. 2023).



Glycine is the only amino acid stable enough to survive a real circuit (Sarvar et al. 2023; Fashu et al. 2026).



It is cheap (about US\$1,700/t, industrial grade) and a permitted food additive; research follows availability, not performance (Fashu et al. 2026).

Glycine wins on durability, not speed: a research-agenda finding, not a deployment one.



Complexing ability toward gold (experimental ranking): asparagine > glycine > histidine > cysteine > phenylalanine > valine > aspartic acid > alanine (Sarvar et al. 2023)



Dissolution of deprotonated acids at high pH (FTIR): cysteine > alanine > glycine > histidine (Tapfuma et al. 2024, as summarized by Fashu et al. 2026). Rankings differ between studies; that is itself the honest finding.



Conditions, scale and independence behind every number on the slides

Every quantitative comparison on the main slides traces to one of these rows

Study	Material	Conditions	Result used on slides	Scale	Independence
Deng et al. 2026	Au–Ag ore 1.41 g/t; Au–Cu ore 2.15 g/t, 1.24% Cu; Cu flotation tailings 2.1 g/t	Bottle-roll, 48 h, 30% solids, pH 10–10.5 (lime), ~25 °C; Jinchan 1–10 kg/t; GlyCat 0.5–4 kg/t each; NaCN 3.5–10 kg/t	Jinchan 100 g/L solution: 7,300 mg/L free CN ⁻ , 12,000 CNO ⁻ , 2,000 Fe(CN) ₆ ⁴⁻ ; leachate 41 mg/L free CN (NaCN 660); Jinchan slowest kinetics; tailings 7% vs 22% (CN) vs 36% (GlyCat)	Laboratory	Independent of Jinchan; GlyCat co-developer
Beyuo et al. 2016	Gold Fields Ghana (Tarkwa) ore	Agitation leach; NaCN 230–250 ppm; Jinchan 200–600 ppm; 24–48 h; pH 11.5	NaCN 97% at 24 h vs Jinchan 91.7% at 400 ppm at 24 h, ~97% only after 48 h; >400 ppm reduced yield; 17 mg/L free CN in Jinchan liquor (102 for NaCN); lime 1.27 vs 1.99 kg/t	Laboratory	Independent
Yilmaz & Sayin 2024	Çukuralan and Kaymaz ores, Turkey	40% solids; 300 mg/L reagent	Jinchan 0.72 kg/t (90.9%) vs NaCN 0.31 kg/t (92.0%); Kaymaz 1.27 vs 0.95 kg/t	Laboratory	Independent
Chaka & Rupprecht 2024	Legadembi carbonaceous ore, Ethiopia (~1.25 g/t)	Bottle-roll 48 h, 50% solids; 9 g/L S ₂ O ₃ , 13.6 g/L NH ₃ , 0.8 g/L CuSO ₄ , pH 10–11.5; NaCN 0.5 g/L	Thiosulfate 91.54% (abstract) / 90.54% (Table 3) vs cyanide 61.70%; no consumption data	Laboratory	Operator-linked author
Redrovan et al. 2025	Guayzimi ore, Zamora Chinchipe, Ecuador (10 g/t Au); not described as ASGM	0.7 M S ₂ O ₃ , 1.75 M glycine, 5 mM Cu, KMnO ₄ every 2 h, pH 9.3, 60 °C, 6 h, 15% solids, 100 g	89.3% in 6 h (cyanide 89.8% in 24 h; ammoniacal thiosulfate 58% in 6 h)	Laboratory	Independent
Tremolada 2018	Peruvian carbonaceous and docile gold ores	Ammonium thiosulfate; stirred tank 120 h; heaps 10,000–20,000 t, 45 d	6.3 kg/t (tank, docile, 93.6%); 7.4 (tank, carbonaceous, 90.8%); 6.5 (heap, docile, 81.2%); 8.6 (pad, carbonaceous, 75.2%)	Pilot / heap	Practitioner, not peer-reviewed
Mohamed et al. 2025	Kombo Laka ASGM tailings, Cameroon (14 g/t Au, 8 g/t Hg)	Cyanide 1.5 g/L NaCN, 24 h; glycine 3 g/L + 3 g/L KMnO ₄ , 10 h; pH 10.5, L/S 3, room temp.	Cyanide 87% vs glycine 45.6%, unequal leach times; at 6 h: 68% vs 29%; Hg removal ≤30% for both	Laboratory	Independent
Barani et al. 2021	Pb–Cu–Zn polymetallic gold ore	Cyanide 200–1,500 ppm ± 0.5–1 M glycine; up to 72 h	500 ppm CN: 46% → 90% with glycine; dose 500 vs 1,500 ppm without glycine (~67% lower), ~80% less cyanide consumed as summarized by Deng et al. 2026; poor selectivity (Cu, Pb, Zn rise)	Laboratory	Independent
Draslovka / Barrick 2024; Draslovka 2023	Bulyanhulu CIL (1.1 Mtpa, ~5.8 g/t, baseline 4.3 kg/t NaCN); Bayan Airag Cu-bearing Au–Ag ore	Commercial CIL after Jan–Mar 2024 pilot; 1,000 t trial heap	80% CN reduction, "comparable" recovery, WAD "undetectable" (no detection limit stated); heap 78% recovery, 65% CN cut	Commercial	Supplier / operator
Nevada Gold Mines 2025	Goldstrike double-refractory carbonaceous ore	Calcium-thiosulfate RIL, 2014–Q1 2023	US\$30 M conversion back to cyanide CIL completed Q1 2023: higher recovery, lower operating cost	Commercial	Operator disclosure (NI 43-101)
Reno et al. 2026	Sumbawa oxide ore, Indonesia (1.55 g/t)	1.5 kg/t reagent, 40% solids, pH 10.5, H ₂ O ₂ to 11 mg/L DO, 24 h	GOLDIX 575 77.5% vs NaCN 70.1%; filtrate free CN 38–70 ppm (GOLDIX) vs 200 (NaCN)	Laboratory	Supplier / operator
Zhang et al. 2022 (compilation, Table 5)	Chinese mine ores and residues (Xiajiadian, Siergou, Hongyuan, Dabaiyang, Paishanlou); Phabon, Laos; Patay, Peru (Sandioss)	Plant and pilot leaches, 17–48 h, pH 10–12.5; Jinchan 0.3–0.9 kg/t (30 kg/t on a roasted residue); carbon or zinc recovery	Jinchan 85–97% vs NaCN 89–97% at equal or lower dose; "cyanide in leach solution <0.004 mg/L" (method not stated); CG505 0.18 mg/L CN in solution, 1.14 mg/L in cake	Plant / pilot (as reported)	Operator- / vendor-linked, Chinese-language primary studies



Source note: Scale: laboratory = kg-scale bottle-roll/agitation; pilot = 10,000–20,000 t heaps; commercial = operating plant. Independence = author affiliation as declared. Last row: vendor-side plant reports compiled by Zhang et al. 2022, for questions on the Chinese literature.

Independent assay protocol for any proprietary "cyanide-free" agent

What "assay first" means in practice: five samples along the circuit, speciated by an independent laboratory

Sample	What it answers	Analytes	Why it matters in a yard
1 • Unused reagent (as sold)	Does the product itself contain or release cyanide on dissolution?	free · WAD · total CN; CNO ⁻ ; Fe(CN) ₆ ; SCN ⁻ ; thiourea; NH ₃ /NH ₄ ⁺ ; metals (Fe, Cu, As, Hg)	Deng et al. 2026 found 7,300 mg/L free CN ⁻ and 2,000 mg/L ferrocyanide in a 100 g/L Jinchuan solution.
2 • Prepared leach solution	What is the working concentration of cyanide species at the dose actually used?	free · WAD · total CN; pH; CNO ⁻ ; SCN ⁻	Sets the exposure level for whoever mixes and doses.
3 • Pregnant solution	Which complex carries the gold, and what else is in solution?	Au, Ag, Cu, Fe, As, Hg; free · WAD · total CN; CNO ⁻ ; SCN ⁻ ; NH ₃	Tells you whether carbon/CIL will work and what the recovery train must handle.
4 • Barren solution	What is discharged or recycled after recovery?	free · WAD · total CN; CNO ⁻ ; SCN ⁻ ; NH ₃ ; COD; metals	Determines destruction/treatment requirement and effluent compliance (free 0.1 · WAD 0.5 · total 1.0 mg/L).
5 • Tailings / residue	What stays on the ground, and how does it behave with mercury?	total CN, WAD CN (leachate test); metals incl. Hg; Hg(CN) ₂ speciation where Hg present	Annex C interface: cyanide-generating agents on mercury-bearing tailings form soluble Hg–CN complexes (Veiga et al. 2020).



Rules of the protocol: sampling and preservation to a cyanide-speciation laboratory (not a colorimetric field kit); silver-nitrate titration on site for free cyanide only; an independent laboratory with no supplier relationship; results published with the program's endorsement decision. Confidential composition means no independent risk assessment.

Research, trade and use, graded by evidence type

Where the gold dressing agents are being tested, sold and used, and how strong the evidence is for each entry

Country / region	What was found	Evidence type	planetGOLD
China	Domestic market leader; named industrial customers (China National Gold, Shandong Gold, Zijin, Zhaojin); plant substitution studies at Xiajiadian, Siergou, Hongyuan and others compiled by Zhang et al. 2022 (Table 5; operator-linked); classed as an ordinary chemical, "no approval needed to buy, store or transport"; eight producers with >250 kt/yr combined capacity.	Regulatory filing / patent	—
Laos · Guinea · Russia	Sole-supply industrial customers named in securities filings: Vangtat Mining (Laos, 3,275–3,640 t/yr); Ningbo Zhongce's Kankan subsidiaries (Guinea, 16,884 t demand 2025); Arktos → Terimalchik mine (Russia, 50 t in 2023 → 1,200 t in 2025). Laos also has operator-reported plant results at Phabon Gold Mine (Guo et al. 2015; Kang et al. 2019, via Zhang et al. 2022).	Regulatory filing / patent	—
Indonesia	Jinchan retails in 1 kg and 25 kg packs on Lazada, Shopee, Tokopedia as "pengganti sianida (non-toxic)"; five named Jakarta distributors; CNFREE marketed against Indonesia's NaCN controls since 2012; GOLDIX laboratory study (Reno et al. 2026, supplier-affiliated) measures free cyanide in its tailings.	Distributor / marketplace	Phase 1
Peru	Equivalent chemistry sold as Gold MAX "lixivante ecológico" (Royal Chemical, Lima; stated LD ₅₀ 50 mg/kg); four university theses validate it at pH 12, 0.5 g/L, 48 h at La Rinconada, Arequipa, Puno, Cusco; marketed to miners on social media. Another GDA (Sandioss) bench-tested on Pataz ore (Heras et al. 2018, via Zhang et al. 2022).	Distributor / marketplace	Phase 1
Kenya	AuLeach GDA marketed for "Kakamega & Migori" beside NaCN: "cyanide-free", clears customs "as standard industrial cargo", avoids Class 6.1 processing at Mombasa, "rapid NEMA EIA approvals". A 2025 NEMA EIA for a Kakamega vat-leach plant specifies conventional NaCN.	Distributor / marketplace	Phase 1
Zimbabwe	CNLITE Africa (Harare) markets CNLITE as "100% cyanide-free", LD ₅₀ >5,000 mg/kg, 96–98% recovery at 0.3–0.6 kg/t; 2026 Mazowe mercury-free trials name hypochlorite and cyanidation, not GDAs; 2019 NAP silent.	Distributor / marketplace	Phase 2
Tanzania	Jinchan marketed for vat leaching; AMGC (c. 2019) investigation of the environmental and health impact of GDA in small-scale mining, likely the only intergovernmental field study; not retrievable.	Intergovernmental (not retrieved)	Phase 2
Ghana	Gold Fields Ghana laboratory comparison (Beyuo et al. 2016): cyanide faster at lower dose (97% vs 91.7% at 24 h; parity only at 48 h); Jinchan liquor 17 mg/L free CN. Vendor (CAMAL) claims TAS96 GDA replaced NaCN at a 300 t/d plant, unaudited.	Peer-reviewed	Phase 2
Honduras	On the vendor's brochure list. Voice memos (Aug 2026): experimental tailings leaching with unnamed lixivants at Macuelizo (Minas y Cuevas concession, legacy heap-leach site); informal miners use cyanide without permits. Reagent unconfirmed; follow up.	Field lead (unconfirmed)	Phase 2
Turkey · Kazakhstan	Independent bench tests (Yilmaz & Sayin 2024; Malyshev et al. 2025): parity on oxide ore, 1.3–2.3× consumption, slower kinetics; official distributor and registered trademark in Kazakhstan (jinchan.kz).	Peer-reviewed	—
Suriname · Guyana · Colombia · Sudan · CAR · Nigeria · Philippines · others	Named on the manufacturer's user-country or brochure lists (27–30+ countries); no distributor, import or regulatory record located. Sudan's Sai Island banned cyanide "mixers" (Nov 2024) without naming GDAs.	Vendor claim	Phase 1–2 / GEF-8



REGULATORY POSITION: no registration, approval, ban, import restriction, seizure or agency statement on Jinchan or any GDA was found in any of the 27 claimed markets; ICMI and the reagent industry have published no position.

Full assessment matrix: 14 criteria

Report 1 §6.1, reproduced without change: ● strong or favorable · ◐ moderate or conditional · ○ weak or unfavorable

Option	1 Recovery	2 Ore fit	3 Process simplicity	4 Cost	5 Supply	6 Worker safety	7 Environ. manageability	8 Detox / effluent	9 Low-skill / no lab	10 Field robustness	11 Regulatory acceptance	12 Scalability	13 ASGM suitability	14 Readiness / maturity
Cyanidation (baseline)	●	●	◐	●	◐	○	◐	◐	◐	◐	●	●	◐	●
Thiosulfate	◐	●	○	◐	◐	◐	◐	◐	○	○	◐	◐	○	◐
Glycine / GlyCat	◐	◐	◐	◐	◐	●	●	◐	○	○	◐	◐	○	◐
Glycine–thiosulfate / CN–glycine hybrid	◐	◐	○	◐	◐	◐	◐	◐	○	○	◐	◐	○	◐
Thiourea	◐	◐	○	○	○	○	○	○	○	○	○	○	○	○
Thiocyanate	◐	◐	○	○	○	◐	◐	○	○	○	○	○	○	○
Polysulfide / alkaline sulfide	◐	◐	○	◐	◐	○	○	○	○	○	○	○	○	○
Halides (Cl / Br / I)	●	◐	○	○	○	○	○	○	○	○	○	○	○	○
Chlorination / hypochlorite	◐	◐	○	◐	◐	○	○	○	○	○	○	◐	○	◐
Cassava / cyanogenic (governed as cyanide)	○	◐	◐	●	●	○	○	○	◐	◐	○	○	◐	○
Proprietary GDAs (Jinchan / CNFREE / CNLITE)	◐	◐	◐	◐	●	○	○	○	○	◐	○	◐	○	○
Novel synthetic (Li-type NESGL)	◐	○	○	○	○	◐	◐	○	○	○	○	○	○	○
Bio-oxidation (pretreatment)	◐	●	○	◐	◐	◐	●	◐	○	○	◐	◐	○	◐
Bioleaching / biogenic	○	◐	○	◐	◐	◐	●	◐	○	○	○	○	○	○
DES / ionic liquids	◐	◐	○	○	○	◐	◐	○	○	○	○	○	○	○

LEGEND

● strong / favorable

◐ moderate / conditional

○ weak / unfavorable

More dots = more favorable for ASGM deployment; the baseline row rates cyanide on the same scale, so ○ on worker safety is weak for cyanide by design. Ratings are screening judgments, not precision scores; criteria and traceability on B6–B8.

Mapping to the seven families on slide 5: sulfur-based = thiosulfate, thiourea, thiocyanate, polysulfide; halide-based = halides (Cl / Br / I), chlorination; amino-acid = glycine / GlyCat; oxidation-assisted or hybrid = the glycine–thiosulfate / CN–glycine hybrid (advanced oxidation processes are not scored in Report 1 §6.1); biological = bio-oxidation, bioleaching; advanced solvent = DES / ionic liquids; proprietary or synthetic (verification category) = cassava / cyanogenic, proprietary GDAs, novel synthetic NESGL; cyanidation is the baseline.

The fourteen assessment criteria and the four-tier classification

Each criterion is a due-diligence question, not a single number; applied identically to every option (Report 1 §5, §6.5)

#	Criterion	What is assessed
1	Metallurgical effectiveness / gold recovery	Achievable recovery on representative feed, not best-case laboratory foil or concentrate.
2	Ore and mineralogical suitability	Free-milling vs refractory; sulfides, copper, preg-robbing carbon.
3	Process complexity and control	Number of reagents; Eh–pH window; oxidant management; monitoring burden.
4	Capital and operating cost	Reagent + oxidant + recovery train + effluent capex/opex; delivered, not headline reagent price.
5	Reagent availability and supply-chain reliability	Local availability, shelf life, transport, storage, import control.
6	Worker health and safety	Acute/chronic toxicity, gas generation, corrosivity, exposure controls.
7	Environmental risk and manageability	Effluent, residues, ecosystem exposure, spill behavior.
8	Detoxification / effluent / waste management	Whether an engineered destruction or recovery step is required and feasible.
9	Technical skills and laboratory requirement	Operator competence and analytical support needed to run it safely.
10	Field robustness	Tolerance of variable feed, weak process control and interruption.
11	Regulatory acceptance	Standing under mining, chemicals and environmental regimes; licensing.
12	Scalability	From village kit to centralized hub without loss of control.
13	Suitability for ASGM conditions	The composite of the above under informal, low-infrastructure operation.
14	Technology readiness and commercial maturity	Laboratory / pilot / demonstration / commercial, and independence of the evidence.



FOUR-TIER ASGM CLASSIFICATION

Tier 1, conditional controlled-pilot candidates: thiosulfate and glycine, in supervised centralized hubs only (Liu et al. 2022; Eksteen et al. 2018; Mohamed et al. 2025).

Tier 2, longer-term R&D: glycine–thiosulfate hybrids, thiocyanate, bio-oxidation as pretreatment, AOPs, DES / ionic liquids (Djokić et al. 2026).

Tier 3, cautionary / proprietary, verify before any use: Jinchan, CNFREE, CNLITE, generic GDAs and the Li-type NESGL; cassava and cyanogenic routes governed as cyanide (Zhang et al. 2022; Guo et al. 2022; Li et al. 2025; Torkaman et al. 2021).

Tier 4, not suitable for dispersed ASGM: thiourea, halides / chlorination, polysulfide, stand-alone bioleaching (Liu et al. 2022).

The tiers are Report 1 §6.5 and broadly match the readiness matrix on slide 7; two placements differ: Report 1 puts thiocyanate and bio-oxidation in Tier 2 (longer-term R&D), whereas slide 7 positions thiocyanate as not yet mature for deployment and bio-oxidation as a hub pretreatment.

Traceability: recovery, ore fit, maturity, cost and supply

Report 1 §6.2 (criteria 1, 2, 14) and §6.3 (criteria 3, 4, 5), with the evidence each rating rests on

Option	Reported recovery	Ore fit	Maturity	Evidence
Thiosulfate	~70–95%, often 80–90% optimized	Carbonaceous, preg-robbing, Cu-bearing	Commercial at one site (Goldstrike) 2014–23; reverted to cyanide Q1 2023 ‡	Djokić et al. 2026; Liu et al. 2022; Xie et al. 2021; Nevada Gold Mines 2025
Glycine / GlyCat	~45% stand-alone on ASGM tailings vs 87% CN; 46% → 90% at 500 ppm CN as CN–glycine hybrid ‡	Cu-gold; alkaline	Commercial (Bulyanhulu) as CN reduction; operator-reported	Mohamed et al. 2025; Barani et al. 2021 (2022 figures unverified); Eksteen et al. 2018
Thiourea	~80–95% laboratory	Some refractory / acid-oxidized	Pilot; 1980s antimony-concentrate circuit (NSW)	Djokić et al. 2026; Borda & Torres 2022; Zhang et al. 2022
Halides	~85–98% on concentrates	Oxidized concentrates	Cl commercial (concentrates); Br / I laboratory	Djokić et al. 2026; Liu et al. 2022
Proprietary GDAs	90.8% on Carlin concentrate (Jinchuan); 85–97% Chinese plant reports; 91.7% at 24 h in Ghana	Concentrate, alkaline / carbon	Marketed; plant use in China (operator-reported); independent laboratory data only (Ghana, Turkey, Curtin) ‡	Guo et al. 2022; Zhang et al. 2022; Beyuo et al. 2016; Deng et al. 2026
Novel synthetic (Li)	87.56% roasted concentrate; 66.15% untreated	Roasted sulfide concentrate	Bench	Li et al. 2025
Bio-oxidation	85–98% with activation (as pretreatment)	Refractory sulfides	Commercial as pretreatment	Djokić et al. 2026; Liu et al. 2022

Option	Indicative reagent cost (US\$/t)	Key cost / control driver	Supply / control note
Cyanide (baseline)	NaCN ~2,700	<5% of processing cost at 0.2–0.5 kg/t ore; mature CIP/CIL recovery	Controlled import; concentrated global supply
Thiosulfate	Na ₂ S ₂ O ₃ ~500	Consumption via degradation erodes the cost edge; needs resin/IX recovery	Widely available
Thiourea	~1,500	High reagent cost + acid handling	Available; carcinogen-controlled
Halides: iodine	I ₂ ~30,000	Reagent cost is the entire obstacle	Specialty supply
Chlorine	Cl ₂ ~250	Cheapest reagent, worst EHS	Gas handling infrastructure required
Polysulfide	Na ₂ S ~400	Cheap, alkaline; H ₂ S risk; high effluent COD	Available
Glycine	Commodity scale (~1,700 industrial grade, 2026 index)	Slow kinetics need oxidant; recyclable (claimed; unvalidated over repeated cycles)	Industrial-scale commodity (Draslovka)

Reagent prices from Liu et al. 2022 (Table S11, after Aylmore 2016) and 2025–26 indices; cost share from JXSC 2025 and Liu et al. 2022. Prices are indicative reagent-only figures and exclude oxidant, recovery-train and effluent capex, the costs that usually decide the outcome.

Traceability: hazards, effluent, ASGM fit, and evidence limitations

Report 1 §6.4 (criteria 6, 7, 8, 13), and the limits of the evidence behind the whole assessment

Option	Principal hazard	Detox / effluent requirement	Net ASGM read
Cyanide	Acute toxicity; HCN gas below pH 9.3; Hg(CN) ₂ on Hg tailings	Engineered destruction (SO ₂ /air, H ₂ O ₂) or recovery	Governable only under stewardship; not for uncontained use
Thiosulfate	Low acute toxicity; ammonia hazard in Cu–NH ₃ system	High-COD effluent needs treatment	Centralized only
Glycine / GlyCat	Low toxicity; contains cyanide by design (GlyCat)	CN-reduced tailings; IX recovers >95% metals, weak Hg uptake	Medium-term, hybrid; not stand-alone
Thiourea	Suspected carcinogen; acidic	Effluent treatment required	Weak fit
Halides / chlorination	Cl ₂ / Br ₂ toxic, corrosive	Gas capture and neutralization	Engineered facility only
Polysulfide	H ₂ S generation on pH excursion	High-COD detox	Unacceptable in an open ASGM setting
Cassava / cyanogenic	Generates CN [–] by hydrolysis	Same as cyanide	Govern as cyanide; not a non-cyanide method
Proprietary GDAs	Cyanide-bearing / -generating; misleading labeling	Presumed same as cyanide; SDS itself prescribes hypochlorite and ferrous sulfate	Verify by assay before any consideration



EVIDENCE LIMITATIONS BEHIND THE ASSESSMENT

Scale: most non-cyanide results are laboratory bottle-roll or agitation tests on kilogram samples; the only pilot data are Peruvian thiosulfate heaps (practitioner report) and the only commercial data are Goldstrike (thiosulfate, reverted) and GlyCat plants (supplier/operator-reported, unaudited).

Independence: the Chinese plant literature on the gold dressing agents is operator- or vendor-linked and does not state cyanide analytical methods; developer-authored glycine papers (Curtin) and the Draslovka/Barrick material are labeled as such.

Comparability: leach times, doses and ores differ between the studies compared (Cameroon glycine 10 h vs cyanide 24 h; Ghana Jinchan 24 vs 48 h); ratings are screening judgments, not a numerical index.

Gaps: no third-party assay of a gold dressing agent as sold; no ecotoxicology; no glycine consumption on ASGM feed; no independent audit of GlyCat plant data; no thiosulfate recovery or effluent treatment at hub scale; residue behavior of every alternative on mercury-bearing tailings untested.

Alternative lixiviants: the key finding in one line

Principal advantage, principal limitation and ASGM implication by technology family



Laboratory or optimized recovery is not equivalent to field performance.



Technical promise is not the same as ASGM suitability.

Technology family	Principal advantage	Principal limitation	ASGM implication
Thiosulfate	Lower toxicity; strongest near-term alternative.	High reagent consumption and solution instability; no plant operating today.	Most credible near-term candidate, but only in controlled hubs.
Glycine / amino acids	Low toxicity; alkaline chemistry; promising for copper-gold ores.	Slower kinetics; oxidant needs; still emerging.	Promising for contained pilots, not direct field deployment.
Thiourea	Fast gold dissolution.	Acidic, unstable and costly; suspected carcinogen.	Technically interesting but weak ASGM fit.
Thiocyanate	Lower acute toxicity than cyanide.	Experimental; acidic and oxidant-intensive.	Not yet mature for deployment.
Halides	Rapid kinetics and strong performance on some concentrates.	Corrosive systems and higher EHS burden.	Only plausible in highly engineered facilities.
Bio-oxidation / biohydrometallurgy	Useful for refractory ores and lower-energy pretreatment.	Slow and operationally demanding.	Best framed as pretreatment rather than direct replacement.
Proprietary gold dressing agents and synthetic lixiviants	Commercial interest and easier retrofit claims.	Cyanide-bearing on independent assay; composition undisclosed.	Use only with independent verification, composition disclosure and cyanide-class controls.
AOPs, deep eutectic solvents and ionic liquids	Interesting emerging research directions.	Scale-up, cost and maturity constraints.	R&D relevance only at present.



KEY TAKEAWAY: no single alternative is ASGM-ready today. Suitability depends on control, monitoring and evidence, not just chemistry.

Where the evidence stops: seven countries, one wall

Every documented case dissolves gold; no ASGM-relevant case has shown recovery from solution, effluent management and delivered cost on ASGM feed

Country case (scale · independence)	Gold dissolved	Recovered from solution	Effluent managed	Cost established	ASGM scale
Ghana: Jinchuan vs NaCN (cyanide faster, at lower dose) Beyuo et al. 2016 · laboratory · independent	SHOWN	NOT SHOWN	NOT SHOWN	NOT SHOWN	NOT SHOWN
Ethiopia: thiosulfate, ~91% on carbonaceous ore Chaka & Rupprecht 2024 · laboratory · operator-linked author	SHOWN	NOT SHOWN	NOT SHOWN	NOT SHOWN	NOT SHOWN
Ecuador: thiosulfate–glycine, 89.3% at 60 °C Redrovan et al. 2025 · laboratory · independent	SHOWN	NOT SHOWN	NOT SHOWN	NOT SHOWN	NOT SHOWN
Peru: ammonium thiosulfate tank / heap, 75–94% Tremolada 2018 · pilot to 20,000 t heaps · practitioner report	SHOWN	PARTIAL	NOT SHOWN	PARTIAL	PARTIAL
Indonesia: GOLDIX (cyanurate-class GDA) vs NaCN; Jinchuan sold in 1 kg retail packs Reno et al. 2026 · laboratory · supplier-affiliated; marketplace listings	SHOWN	NOT SHOWN	PARTIAL (detox tested)	NOT SHOWN	RETAIL USE, UNDOCUMENTED
Kenya: GDA marketed into Kakamega / Migori to avoid Class 6.1 controls Distributor page (United Chemical 2026); no documented use	?	?	?	?	?
Honduras: field lead, tailings leaching with unnamed lixivants, Macuelizo planetGOLD Honduras voice memos, Aug 2026 · reagent unconfirmed	?	NOT SHOWN	NOT SHOWN	NOT SHOWN	?
Mongolia: GlyCat trial heap, LSM analogue Draslovka 2023 · 1,000 t trial · supplier-reported	SHOWN	SHOWN	PARTIAL	SHOWN	NOT ASGM



THE WALL: no ASGM-relevant case gets past recovery from solution. Two countries are field leads or marketing only; five are documented, all laboratory or pilot. Only the large-scale-mining analogue crosses the wall, on supplier-reported data.

What must be in place, whatever the lixiviant

Four controls for each audience; every item applies to cyanide today and to any reagent that follows it



OPERATORS

Pre-concentrate first

gravity before any leach; cut the mass needing chemistry by an order of magnitude (Gunson & Veiga 2020)

Contained circuit

alkaline, lined, no open discharge; carbon or resin recovery specified before the first ton

Reagent custody

licensed supply, locked storage, dose by titration, for ANY lixiviant, branded or not (Stapper, Dales et al. 2021)

People

PPE, training and rotation for whoever touches the pulp, women and children included



REGULATORS

License the reagent, not the brand

Jinchan, CNFREE, GlyCat and cassava fall under cyanide rules; "ordinary cargo, no restrictions" is a red flag; watch imports under HS 3824.99 "chemical preparations", not only HS 2837 "cyanides"

Set the numbers

free 0.1 · WAD 0.5 · total 1.0 mg/L in effluent (World Bank 1995); WAD below 50 mg/L in any open pond; and a laboratory that can measure them

Annex C, Minamata

no cyanide, or cyanide-generating agent, on tailings to which mercury has been added without first removing it (1(b)(iv)); the bridge to Presentation 2

Inspect the hub, not the hillside

permit and monitor centralized processing plants, where control is possible



SUPPORT PROGRAMS

Assay before you endorse

independent five-sample speciation of any proprietary agent before it enters a program

Centralized, supervised plants

the model the WGC, Ghana and the AGC committed to in July–August 2026 (WGC Gold Processing Initiative)

Borrow the Code's architecture

independent verification, public reporting, corrective-action loops, not its fee model (ICMI 2026)

Demonstrate the whole flowsheet

leach + recovery + effluent + delivered cost, on real ASGM feed, before scaling anything



The assurance model: the Code showed that independent verification, transparency and corrective-action loops can civilize a hazardous reagent in the formal sector. A simplified, subsidized analogue is what reaches ASGM (Stapper, Dales, Velasquez & Keane 2021; Report 1, Cyanide Code and implementation sections).



Source note: Gunson & Veiga 2020; Stapper, Dales, Velasquez & Keane 2021; World Bank 1995 (mine effluent guidelines); Minamata Convention 2013, Annex C 1(b)(iv); World Gold Council 2026 (Gold Processing Initiative MoUs); ICMI 2026; Guangxi Senhe listing documents (customs classification).

What to do now, and what we still need to know



PRACTICAL RECOMMENDATIONS

- **Operators:** gravity pre-concentration before any leach; contained alkaline circuit with carbon or resin recovery specified; titration-based dosing; PPE, training and rotation for pulp handlers.
- **Regulators:** license reagent chemistry, not brands, and watch imports under HS 3824.99 as well as HS 2837; adopt free/WAD/total cyanide effluent limits and fund a speciation laboratory; permit and inspect centralized plants; apply Annex C to cyanide-generating agents.
- **Support programs:** require the five-sample independent assay before endorsing any proprietary agent; invest in centralized, supervised plants; replicate the Code's verification and corrective-action loops in a subsidized form.
- **Sequence:** responsible cyanidation first; pilot alkaline glycine or GlyCat in a hub only with the whole flowsheet (leach + recovery + effluent + delivered cost) demonstrated on real ASGM feed.
- **Communicate:** treat "cyanide-free", "non-toxic" and "ordinary cargo" as claims to be tested, and say so publicly; put the labeling question formally to ICMI and to the reagent industry; neither has a published position.



PRIORITY EVIDENCE GAPS: FUTURE RESEARCH

- **Proprietary agents:** assay of retail-purchased Jinchan, CNFREE and CNLITE at an accredited laboratory (free/WAD/total cyanide, ferrocyanide, cyanate, thiourea); no published third-party analysis of product as sold exists; reconcile the three circulating recipes; ecotoxicology; residue behavior on mercury-bearing tailings.
- **Alkaline glycine on ASGM feed:** recovery from solution on site carbon, glycine and oxidant consumption per ton, recyclability over repeated cycles, delivered cost, at matched leach times against cyanide.
- **GlyCat:** independent audit of the Bulyanhulu and Bayan Airag data; a small-scale hub configuration trial with WAD-cyanide monitoring.
- **Thiosulfate:** resin recovery and ammonia/COD effluent treatment at pilot-hub scale on carbonaceous ASGM ore; consumption and cost on ASGM feed; lessons from the Goldstrike reversal.
- **Exposure and field use:** cyanide speciation in water, soil and biomonitoring at ASGM processing sites; confirm GDA use in program countries (Honduras/Macuelizo lead; Indonesia, Kenya, Peru, Tanzania, Zimbabwe; the AMGC Tanzania report); gender-disaggregated exposure data.
- **Mercury interface:** behavior of every alternative on mercury-contaminated tailings: the question Presentation 2 takes up.



PRIORITY ACTIONS: ore characterization, gravity-first recovery, mercury screening, controlled piloting, and independent verification of proprietary products.



Source note: Next: Presentation 2, How clean is clean? Removing mercury from ASGM tailings (Paul Cordy, Cordy Geosciences). Report 1 (draft 3, July 2026) carries the full 14-criterion matrix and the annotated evidence base; all supplier figures are labeled as such.

Where alternative technologies fit in an ASGM transition pathway

A technology positioning ladder showing increasing technical and management complexity



References with DOIs and stable links

Titles as published; evidence-type annotations are given in the speaker notes of each slide

African Minerals and Geosciences Centre (c. 2019; A. Mkama). Report on investigation of environmental and health impact of the gold dressing agent (GDA) used for gold extraction. <https://www.researchgate.net/publication/334132538>

Aylmore, M.G. (2016). Alternative lixiviants to cyanide for leaching gold ores. In: Adams, M.D. (ed.), *Gold Ore Processing*, 2nd ed., Elsevier, pp. 447–484. <https://doi.org/10.1016/B978-0-444-63658-4.00027-X>

Aylmore, M.G. & Muir, D.M. (2001). Thiosulfate leaching of gold — a review. *Minerals Engineering* 14(2):135–174. [https://doi.org/10.1016/S0892-6875\(00\)00172-2](https://doi.org/10.1016/S0892-6875(00)00172-2)

Barani, K., Dehghani, M., Azadi, M.R. & Karrech, A. (2021). Leaching of a polymetal gold ore and reducing cyanide consumption using cyanide-glycine solutions. *Minerals Engineering* 163:106802. <https://doi.org/10.1016/j.mineng.2021.106802>

Barani, K., Kogani, Y. & Nazarian, F. (2022). Leaching of complex gold ore using a cyanide-glycine solution. *Minerals Engineering* 180:107475. <https://doi.org/10.1016/j.mineng.2022.107475>

Barrick Gold (2024). North Mara & Bulyanhulu site visit presentation, May 2024, slide 32 "Glycine project and benefit". https://s25.q4cdn.com/322814910/files/doc_presentations/2024/05/North-Mara_Bulyanhulu_Site_Visit.pdf

Beyuo, M., Abaka-Wood, G.B., Asamoah, R.K., Kabenlah, A. & Amankwah, R.K. (2016). A comparative study of sodium cyanide and Jinchan™ gold leaching reagents: a case study at Goldfields Ghana Limited. 4th UMaT Biennial International Mining and Mineral Conference, Tarkwa, pp. 195–199. <https://www.researchgate.net/publication/311440298>

Borda, J. & Torres, R. (2022). Prospects for thiourea as a leaching agent in Colombian gold small-scale mining. *Journal of Sustainable Mining* 21. <https://doi.org/10.46873/2300-3960.1364>

Chaka, K.J. & Rupprecht, S.M. (2024). Thiosulfate leaching in carbonaceous gold-bearing ores in Ethiopia. *Scientific Reports* 14. <https://doi.org/10.1038/s41598-024-69646-3>

CNLITE Africa / Altitude Groupe (Pvt) Ltd, Harare. Products page: CNLITE "100% cyanide-free", "non-toxic (LD50 >5,000 mg/kg)", 96–98% recovery at 0.3–0.6 kg/t. <https://cnliteafrica.com/products>

Dales, K. & Cordy, P. (2026). Chemical Alternatives to Cyanide in ASGM. planetGOLD/NRDC Report 1, Revised Draft 3, 21 July 2026 (for internal review).

De Tomi, G. et al. (2025). From agro-industrial waste to gold lixiviant: evaluating cassava wastewater applications in artisanal mining. *Mining* 5(4):64. <https://doi.org/10.3390/mining5040064>

Deng, Z., Schwehr, B.J., Yang, A. & Oraby, E.A. (2026). A comparative study of cyanide, Jinchan, and GlyCat™ for gold ore leaching. *Hydrometallurgy* 244:106799. <https://doi.org/10.1016/j.hydromet.2026.106799>

Djokić, J., Nikolić, S., Dimitrijević, S., Zhong, S. & Kamberović, Z. (2026). Gold recovery beyond ores: sources, processes, challenges, and prospects. *Metals* 16:595. <https://doi.org/10.3390/met16060595>

Dong, Z., Jiang, T., Xu, B., Yang, Y. & Li, Q. (2017). Recovery of gold from pregnant thiosulfate solutions by the resin adsorption technique. *Metals* 7(12):555. <https://doi.org/10.3390/met7120555>

Draslovka (2023). First global commercial heap leach use of GlyCat™ process — Bayan Airag, Mongolia. Press release, 29 Aug 2023. <https://www.businesswire.com/news/home/20230829435585/en/>

Draslovka & Barrick Gold (2024). Draslovka and Barrick Gold commence commercial use of glycine leaching technology (GlyCat™) at Bulyanhulu — press release, 29 Feb 2024, and Draslovka case study "GlyCat™ enables 80% reduction in CN & detoxification". https://www.draslovka.com/content/files/downloads/GlyCat_Case_Study_Barrick_Bulyanhulu.pdf

Eksteen, J., Oraby, E., Tanda, B. et al. (2018). Towards industrial implementation of glycine-based leach and adsorption technologies for gold-copper ores. *Canadian Metallurgical Quarterly* 57:390–398. <https://doi.org/10.1080/00084433.2017.1391736>

Fashu, S., Mukuya, A. & Kanhukamwe, Q. (2026). On the use of amino acids to leach precious metals from primary and secondary resources. *Metals* 16(7):724. <https://doi.org/10.3390/met16070724>



Source note: Reference register of the presentation; the same entries appear, with evidence-type annotations, in the notes of the slides that cite them.

Full references with DOIs and stable links

Titles as published; evidence-type annotations are given in the speaker notes of each slide

Global Initiative Against Transnational Organized Crime (2022). Sodium cyanide in Kenya's gold market. Risk Bulletin 24. <https://riskbulletins.globalinitiative.net/esa-obs-024/01-sodium-cyanide-in-kenyas-gold-market.html>

Global Market Insights (2026). Sodium cyanide market size and forecast 2026–2035 (March 2026). <https://www.gminsights.com/industry-analysis/sodium-cyanide-market>

Guangxi Senhe High Technology Co., Ltd (2024–2026). Beijing Stock Exchange listing application documents, Kaiyuan Securities note and Zhong Lun supplementary legal opinion (Dec 2024): product class 环保类氰药剂 ("environment-friendly cyanide-class reagent"); raw materials soda ash, urea, sodium cyanate; overseas customers in Laos, Guinea and Russia; "no approval needed to buy, store or transport". https://m.sohu.com/a/1055294910_121123885 ; <http://file.finance.sina.com.cn/211.154.219.97:9494/MRGG/SBGG/2024/2024-12/2024-12-31/10685035.PDF>

Gunson, A.J. & Veiga, M.M. (2020). Gravity concentration in artisanal gold mining. Minerals 10(11):1026. <https://doi.org/10.3390/min10111026>

Guo, X., Liu, Z., Tian, Q., Li, D. & Zhang, L. (2022). Gold extraction from Carlin-type concentrate by a novel environmentally friendly lixiviant. Hydrometallurgy 211:105884. <https://doi.org/10.1016/j.hydromet.2022.105884>

Hilson, G. & Monhemius, A.J. (2006). Alternatives to cyanide in the gold mining industry: what prospects for the future? Journal of Cleaner Production 14(12–13):1158–1167. <https://doi.org/10.1016/j.jclepro.2004.09.005>

Hinton, J.J., Veiga, M.M. & Beinhoff, C. (2003). Women and artisanal mining: gender roles and the road ahead. In: Hilson, G.M. (ed.), The Socio-Economic Impacts of Artisanal and Small-Scale Mining in Developing Countries. A.A. Balkema, Lisse.

INCAS Consulting for planetGOLD (2026). Jinchan gold dressing agent: voice-memo summary and global evidence review. Technical note, Report 1 input, 10 September 2026 (draft for internal review)

Indicative reagent prices: NaCN — IMARC Q1–Q2 2026 (Argentina US\$2,358–2,408/t; Brazil ~2,140; Australia ~2,100; USA ~3,250; Indonesia ~1,740) <https://www.imarcgroup.com/sodium-cyanide-price-trend> ; glycine industrial grade — ChemAnalyst Q2 2026 (USA US\$1,734/t; China 1,530) <https://www.chemanalyst.com/Pricing-data/glycine-1571> ; sodium thiosulfate — IMARC Q4 2025 (US\$475–890/t by region) <https://www.imarcgroup.com/sodium-thiosulfate-pricing-report>

International Cyanide Management Institute (2026). The Code newsletter Q1 2026: 323 certified operations in 46 countries; 248 signatory companies at end-2025. <https://cyanidecode.org/wp-content/uploads/2026/04/Newsltr1Q2026.pdf>

Johnson, C.A., Leinz, R.W., Grimes, D.J. & Rye, R.O. (2002). Photochemical changes in cyanide speciation in drainage from a precious metal ore heap. Environmental Science & Technology 36(5):840–845. <https://pubs.usgs.gov/publication/70024770>

Jorjani, E. & Sabzkoohi, H.A. (2021). Gold leaching from ores using biogenic lixiviants: a review. Hydrometallurgy 203:105644. <https://doi.org/10.1016/j.hydromet.2021.105644>

JXSC (2025). Understanding the gold cyanide extraction process. Technical note, accessed 2026.

Kohio, E.N., Sossou, S.K., Karoui, H. & Yacouba, H. (2025). Environmental cyanide pollution from artisanal gold mining in Burkina Faso: human exposure risk analysis based on a conceptual site model. International Journal of Environmental Research and Public Health 22(7):1125. <https://doi.org/10.3390/ijerph22071125>

Korobushkina, E.D., Karavaiko, G.I. & Korobushkin, I.M. (1983). Biochemistry of gold. Ecological Bulletins 35:325–333.

Kuyucak, N. & Akcil, A. (2013). Cyanide and removal options from effluents in gold mining and metallurgical processes. Minerals Engineering 50–51:13–29. <https://doi.org/10.1016/j.mineng.2013.05.027>

La Brooy, S.R., Linge, H.G. & Walker, G.S. (1994). Review of gold extraction from ores. Minerals Engineering 7(10):1213–1241. [https://doi.org/10.1016/0892-6875\(94\)90114-7](https://doi.org/10.1016/0892-6875(94)90114-7)

Li, H., Sun, H., Ye, G., Que, X., Wang, Y., Chen, Y. & Chen, B. (2018). Treatment of Xiajiadian gold mine with environment-friendly reagent Jinchan as a substitute for sodium cyanide. Gold Science and Technology 26(5):682–688. <https://doi.org/10.11872/j.issn.1005-2518.2018.05.682>

Li, J., Sun, C., Kou, J., Wang, P. & Liu, X. (2025). Development of a gold leaching reagent as an alternative to cyanide: synthesis and performance evaluation. International Journal of Minerals, Metallurgy and Materials 32(4):835–850. <https://doi.org/10.1007/s12613-024-2957-x>



Full references with DOIs and stable links

Titles as published; evidence-type annotations are given in the speaker notes of each slide

Liu, Z.-W., Guo, X.-Y., Tian, Q.-H. & Zhang, L. (2022). A systematic review of gold extraction: fundamentals, advancements, and challenges toward alternative lixiviants. *Journal of Hazardous Materials* 440:129778. <https://doi.org/10.1016/j.jhazmat.2022.129778>

Malone, A. et al. (2023). Transitional dynamics from mercury to cyanide-based processing in artisanal and small-scale gold mining: social, economic, geochemical, and environmental considerations. *Science of the Total Environment* 898:165492. <https://doi.org/10.1016/j.scitotenv.2023.165492>

Malyshev, V., Karimova, L., Sagatbek, S., Kharchenko, Y. & Magaz, A. (2025). A comparative study of gold leaching kinetics using alternative reagents from concentrates of low-grade ores. *Minerals* 15(3):282. <https://doi.org/10.3390/min15030282>

Marsden, J.O. & House, C.I. (2006). *The Chemistry of Gold Extraction*, 2nd ed. SME, Littleton CO. ISBN 978-0-87335-240-6

Minamata Convention on Mercury (2013), Annex C, para. 1(b)(iv): actions to eliminate "cyanide leaching in sediment, ore or tailings to which mercury has been added without first removing the mercury". <https://digitallibrary.un.org/record/758711/files/MinamataConventiontextEn.pdf>

Mohamed, A.H.A., Oupoungam, A.A., Boyraz, T., Alp, İ. & Sinon, A. (2025). Characterization and evaluation of glycine and cyanide leaching application on artisanal gold mining tailings. *Physicochemical Problems of Mineral Processing* 61(3):205177. <https://doi.org/10.37190/ppmp/205177>

Mordor Intelligence (2026). Sodium cyanide market — size, share and forecast (page updated 30 July 2026). <https://www.mordorintelligence.com/industry-reports/sodium-cyanide-market>

Nevada Gold Mines (2025). NI 43-101 Technical Report on the Carlin Complex, effective 31 Dec 2024, pp. 20–21: "A \$30M conversion from the calcium thiosulphate RIL process to standard cyanide leaching with CIL was completed in Q1 2023." https://s25.q4cdn.com/322814910/files/doc_downloads/ngm/Carlin_Complex_TechnicalReport_14032025.pdf

Ondayo, M.A., Watts, M.J., Hamilton, E.M., Mitchell, C., Mankelow, J. & Osano, O. (2023). Artisanal gold mining in Kakamega and Vihiga counties, Kenya: potential human exposure and health risk. *Environmental Geochemistry and Health* 45(8):6543–6565. <https://doi.org/10.1007/s10653-023-01647-z>

Oraby, E.A. & Eksteen, J.J. (2015). The leaching of gold, silver and their alloys in alkaline glycine–peroxide solutions and their adsorption on carbon. *Hydrometallurgy* 152:199–203. <https://doi.org/10.1016/j.hydromet.2014.12.015>

Oraby, E.A., Eksteen, J.J. & O'Connor, G.M. (2020). Gold leaching from oxide ores in alkaline glycine solutions in the presence of permanganate. *Hydrometallurgy* 198:105527. <https://doi.org/10.1016/j.hydromet.2020.105527>

Orica Limited (2024). Acquisition of Cyanco — ASX announcement, 21 Feb 2024 (US\$640 M enterprise value; completed 30 Apr 2024). <https://announcements.asx.com.au/asxpdf/20240221/pdf/060m19s19jjs5n.pdf>

Redrovan, A.S., de la Torre, E. & Aragón-Tobar, C.F. (2025). Gold leaching from an auriferous ore by alkaline thiosulfate–glycine–copper solution. *Metals* 15(2):204. <https://doi.org/10.3390/met15020204>

Reno, F., Aulia, R.H. & Dian, P.N. (2026). Sodium cyanide replacement with environmentally friendly polymerized cyanide GOLDIX for leaching of gold ore from Sumbawa, Indonesia. *E3S Web of Conferences* 693:02007. <https://doi.org/10.1051/e3sconf/202669302007>

Royal Chemical (Lima). Gold MAX® lixiviante ecológico — ficha técnica (Na₂O 35–50%, N 12–20%; stated LD50 50 mg/kg); university theses UNICA 2022 (La Rinconada), UNSA Arequipa, UNAP Puno, UNSAAC Cusco validating use at pH 12, 0.5 g/L, 48 h. <https://repositorio.unica.edu.pe/items/e1f5bf52-d1fb-4312-a577-34f2a318d3e1>

Sarvar, M., Shafaei Tonkaboni, Z., Noaparast, M., Badiei, A. & Amiri, A. (2023). Application of amino acids for gold leaching: effective parameters and the role of amino acid structure. *Journal of Cleaner Production* 391:136123. <https://doi.org/10.1016/j.jclepro.2023.136123>

Shenzhen Toby Laboratories (2024). Safety data sheet TBR-C-202402-0120-1 for Jinchan gold dressing agent, "prepared based on the information provided by client"; elemental breakdown only (C 22.0%, Na 38.3%, N 15.4%, O 22.9%, Fe 0.96%, Cl 0.36%); declares product outside GB 12268, IMDG and IATA; prescribes calcium hypochlorite and ferrous sulfate for wastewater

Stapper, D., Dales, K., Velasquez, P. & Keane, S. (2021). Best Management Practices for Cyanide Use in the Small-Scale Gold Mining Sector. planetGOLD Programme (GEF/UNEP), December 2021. <https://www.planetgold.org/best-management-practices-cyanide-use-small-scale-gold-mining-sector>

Surimbayev, B., Bolotova, L., Akcil, A. & Ghorbani, Y. (2026). Gold mining tailings as a sustainable secondary resource: environmental impacts, processing technologies and metal recovery. A review. *Mineral Processing and Extractive Metallurgy Review*. <https://doi.org/10.1080/08827508.2026.2693551>



Source note: Reference register of the presentation; the same entries appear, with evidence-type annotations, in the notes of the slides that cite them.

Full references with DOIs and stable links

Titles as published; evidence-type annotations are given in the speaker notes of each slide

Tapfuma, A., Akdogan, G., Luckay, R.C. & Tadie, M. (2024). Investigation of gold–amino acid interactions using FTIR spectroscopy and correlation with gold dissolution: studies at different pH values. *Minerals Engineering* 209:108633. <https://doi.org/10.1016/j.mineng.2024.108633>

Tauetsile, P.J., Oraby, E.A. & Eksteen, J.J. (2019). Activated carbon adsorption of gold from cyanide-starved glycine solutions containing copper. Part 1: Isotherms. *Separation and Purification Technology* 211:594–601. <https://doi.org/10.1016/j.seppur.2018.09.024>

Teng, P., Huang, Q., Long, Y. & Fang, M. (2010/2012). Mineral-dressing agent for noble metal, preparation method and application thereof. Patent WO2012079283A1 / CN102321807B; priority CN201010589648.9, 15 Dec 2010 — the patent number ZL201010589648.9 printed on Jinchuan packaging (INCAS/planetGOLD 2026). <https://patents.google.com/patent/WO2012079283A1/en>

Torkaman, P., Veiga, M.M., de Andrade Lima, L.R.P. et al. (2021). Leaching gold with cassava: an option to eliminate mercury use in artisanal gold mining. *Journal of Cleaner Production* 311:127531. <https://doi.org/10.1016/j.jclepro.2021.127531>

Tremolada, J.C. (2018). Lixiviación de minerales refractarios de oro con tiosulfato de amonio. Instituto de Ingenieros de Minas del Perú, 30 May 2018. <https://onemine.org/documents/lixiviaci-n-de-minerales-refractarios-de-oro-con-tiosulfato-de-amonio-poderosa-tecnolog-a-de-reemplazo-del-cianuro-en-operaciones-mineras-a-gran-escala>

United Chemical Co. Ltd (2026). Kenya gold mining chemicals: sodium cyanide, activated carbon and AuLeach™ GDA for Kakamega and Migori — distributor web page ("avoids the agonizingly slow Class 6.1 dangerous goods processing"; "rapid NEMA EIA approvals"). <https://www.unitedchemicalcn.com/kenya-Sodium-Cyanide-activated-carbon/kenya-gold-mining-chemicals.html>

US FDA, 21 CFR §172.812 (glycine — permitted food additive; note §170.50: no longer GRAS for human food); EU Regulation (EC) 1333/2008, E 640 glycine and its sodium salt. Supplier SDS (e.g. Redox 2023): "Not regulated for land/sea/air transport" — no UN number.

Veiga, M.M., Kiefer, A.M. et al. (2020). Reaction of cyanide with Hg0-contaminated gold mining tailings produces soluble mercuric cyanide complexes. *Chemical Research in Toxicology* 33(11):2834–2844. <https://doi.org/10.1021/acs.chemrestox.0c00211>

Verbrugge, B., Lanzano, C. & Libassi, M. (2021). The cyanide revolution: efficiency gains and exclusion in artisanal- and small-scale gold mining. *Geoforum* 126:267–276. <https://doi.org/10.1016/j.geoforum.2021.07.030>

World Bank (1995). Industrial Pollution Prevention and Abatement — mine effluent guidelines: free CN 0.1 mg/L, WAD CN 0.5 mg/L, total CN 1.0 mg/L; wildlife protection where WAD CN >50 mg/L in open waters (as cited in Stapper, Dales, Velasquez & Keane 2021)

World Gold Council (2026). Gold Processing Initiative — MoUs with the Government of Ghana (21 Jul 2026), OCIM (5 Aug 2026), Dynacor (10 Aug 2026) and the Artisanal Gold Council (12 Aug 2026) on centralized processing plants for responsible ASGM. <https://www.gold.org/news-and-events/press-releases>

Xie, F., Chen, J.-N., Wang, J. & Wang, W. (2021). Review of gold leaching in thiosulfate-based solutions. *Transactions of Nonferrous Metals Society of China* 31. [https://doi.org/10.1016/S1003-6326\(21\)65745-X](https://doi.org/10.1016/S1003-6326(21)65745-X)

Yarar, B. (2002). Long term persistence of cyanide species in mine waste environments. *Tailings and Mine Waste '02*, Swets & Zeitlinger, p. 197

Yin, W. & Miao, Y. (2019). Research status and application progress of environment-friendly leaching agents for gold ores. *Conservation and Utilization of Mineral Resources* 39(1) (in Chinese). <https://html.rhhz.net/KCBHYLY/html/dd303519-415e-4017-b911-b065dab6eb20.htm>

Yilmaz, H. & Sayin, Z.E. (2024). Effect of sodium cyanide and Jin Chan chemicals on gold gain from ore. *Journal of Polytechnic* 27(4):1387–1397. <https://doi.org/10.2339/politeknik.1259655>

Zhang, Y., Cui, M., Wang, J., Liu, X. & Lyu, X. (2022). A review of gold extraction using alternatives to cyanide: focus on current status and future prospects of the novel eco-friendly synthetic gold lixiviants. *Minerals Engineering* 176:107336. <https://doi.org/10.1016/j.mineng.2021.107336>





Making a world of difference
in small-scale gold mining.

A GEF Initiative

How clean is clean? Removing mercury from ASGM Tailings

Paul Cordy

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Annex C

(b) Actions to eliminate:

(iv) Cyanide leaching in sediment, ore or **tailings** to which mercury has been added **without first removing the mercury;**

Practical mercury removal from tailings

- ▶ Minamata compliance demands Hg removal with little guidance
 - ▶ Cyanidation is exploding, Hg contaminated tails are ubiquitous
 - ▶ As with gold mining, multi-stage separation
 - ▶ Source: Journal articles, NGO reports, Expert interviews
-



Practical mercury removal from tailings

- ▶ All old technologies, with new research into ASGM field application
- ▶ Clay rich, old acid generating tailings reduce recovery
- ▶ Typical tailings have 100-1000 ppm of Hg or more, 200-300 ppm or less is more usual

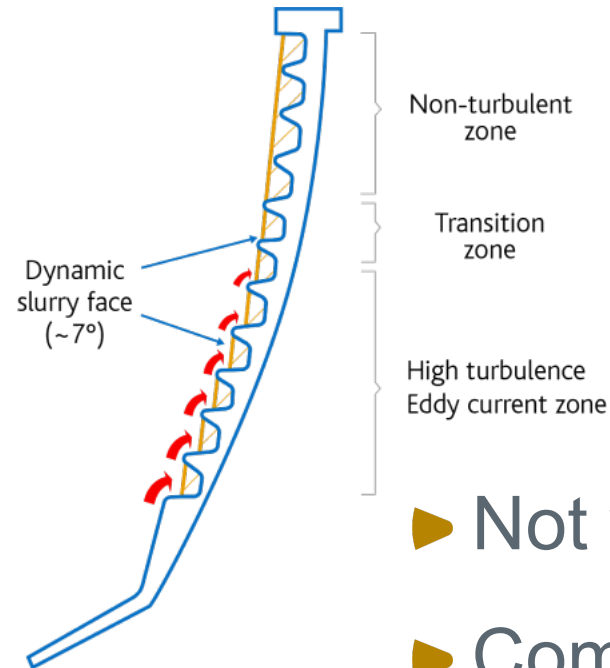


Assessment framework

- ▶ Deployment readiness
- ▶ Effectiveness/recovery (% range)
- ▶ Tailings compatibility
- ▶ Cost
- ▶ Equipment/reagent availability & supply chain
- ▶ Ease of adoption at ASGM scale
- ▶ Health & environmental risk
- ▶ Gender dimensions
- ▶ Evidence confidence
- ▶ Governance and implementation capacity



Gravity concentration



- ▶ Not very efficient <15%
- ▶ Common in gold plants
- ▶ New centrifuge design more effective
- ▶ Costly relative to recovery



Gravity

Deployment Readiness	Widely used for gold and mercury
Recovery (% range)	< 15% for mercury
Tailings compatibility	Universal
Estimated cost (\$/T)	ASGM accessible
Availability	Broadly available
Ease of adoption	Easily adopted, artisanal scale proven
Health & env risk	Low, chemical free
Gender dimensions	High cost relative to recovery
Evidence confidence	Robust history of field implementation
Governance and implementation	Possible anywhere energy and space is available.

Foam floatation

- ▶ Historic, proven, widespread and relatively financially accessible process in mining
 - ▶ Routinely used for extraction of gold and other fine grained hydrophobic minerals like sulphides
 - ▶ Fatty acids and sulfur-based organics (Xanthate), stick to gold, mercury and the foam
 - ▶ Mercury recovery values of up to 75% and over 90% silver for gold and silver can be recovered by this method
-



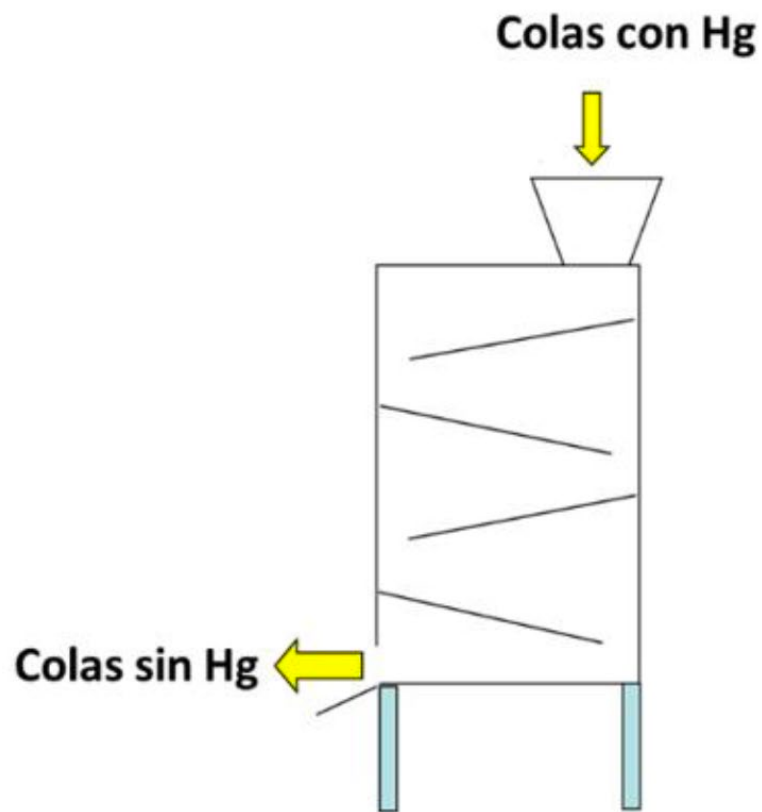


Foam floatation

Deployment readiness	Widely used for gold
Recovery (% range)	75-99%
Tailings compatibility	Universal, great with sulphides
Estimated cost (\$/T)	Modest cost
Availability	Common in ore processing plants.
Ease of adoption	Easily adopted, especially, with small scale cyanidation
Health & env risk	Non toxic exist, toxic = efficient
Gender dimensions	No apparent use case
Evidence confidence	Ubiquitous industrial mining process
Gov+ implementation capacity	Allowed in most countries. Scalable



Copper-Silver plates



- ▶ cheap, effective, and easy to build in ASGM context
- ▶ Up to 85% efficiency was possible using real Colombian tailings
- ▶ Lower recovery from older acid generating tailings
- ▶ Commercially successful in Costa Rica
- ▶ Scalable, reactivate with vinegar

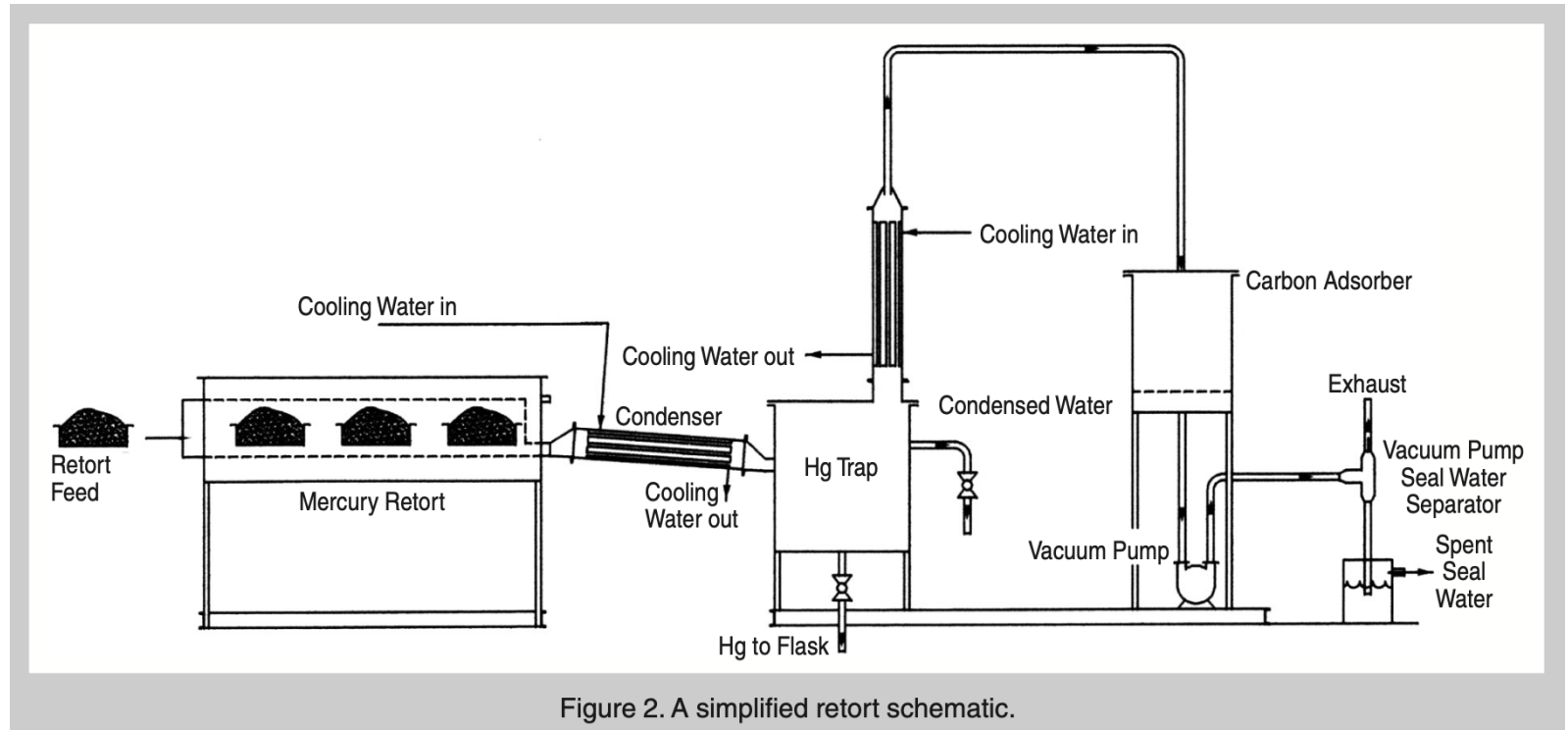
Figure 3 Silver-Covered Copper Plates in a ZigZag Arrangement



Copper /silver plates

Deployment readiness	Field tested and widely applicable
Recovery (% range)	60-95%
Tailings compatibility	Old acidic or muddy tailings are bad
Estimated cost (\$/T)	Accessible to anyone
Availability	Ubiquitous in cyanide operations
Ease of adoption	Easily adopted and locally produced
Health & env risk	Low, if electroplating carefully
Gender dimensions	Naturally down-scalable, transportable
Evidence confidence	Robust lab and field research
Gov+ implementation capacity	Best in formal cyanidation facilities. Scalable

Thermal desorption (retort)



Washburn et al. 2003

- ▶ Field tested and widely used for gold
- ▶ Easily adopted, proven artisanal and industrial designs
- ▶ 90-99% efficient
- ▶ Also can liberate more gold by degrading sulphides
- ▶ 400 and 600 °C under 0.20 to 0.25 atmosphere



Thermal desorption (retort)

Deployment readiness	Ancient, simple, proven, scalable
Recovery (% range)	90-99%
Tailings compatibility	Universal, also liberates gold
Estimated cost (\$/T)	500 -1000 USD per tonne
Availability	Broadly available, common in soil remediation and mining.
Ease of adoption	Easily adopted, artisanal designs proven
Health & env risk	Some fugitive mercury vapour
Gender dimensions	Locally built artisanal solutions
Evidence confidence	Widely used for roasting amalgam, sulphides, soil decontamination.
Gov + implementation capacity	Can meet emissions and disposal standards

Artisanal Zinc retort

Propane heated, water cooled

- ▶ Eliminates open air burning of gold and mercury loaded zinc
- ▶ Water cooled, vapour sealed by submersion of bottom edge





Solar retort

Navarro et al. 2009

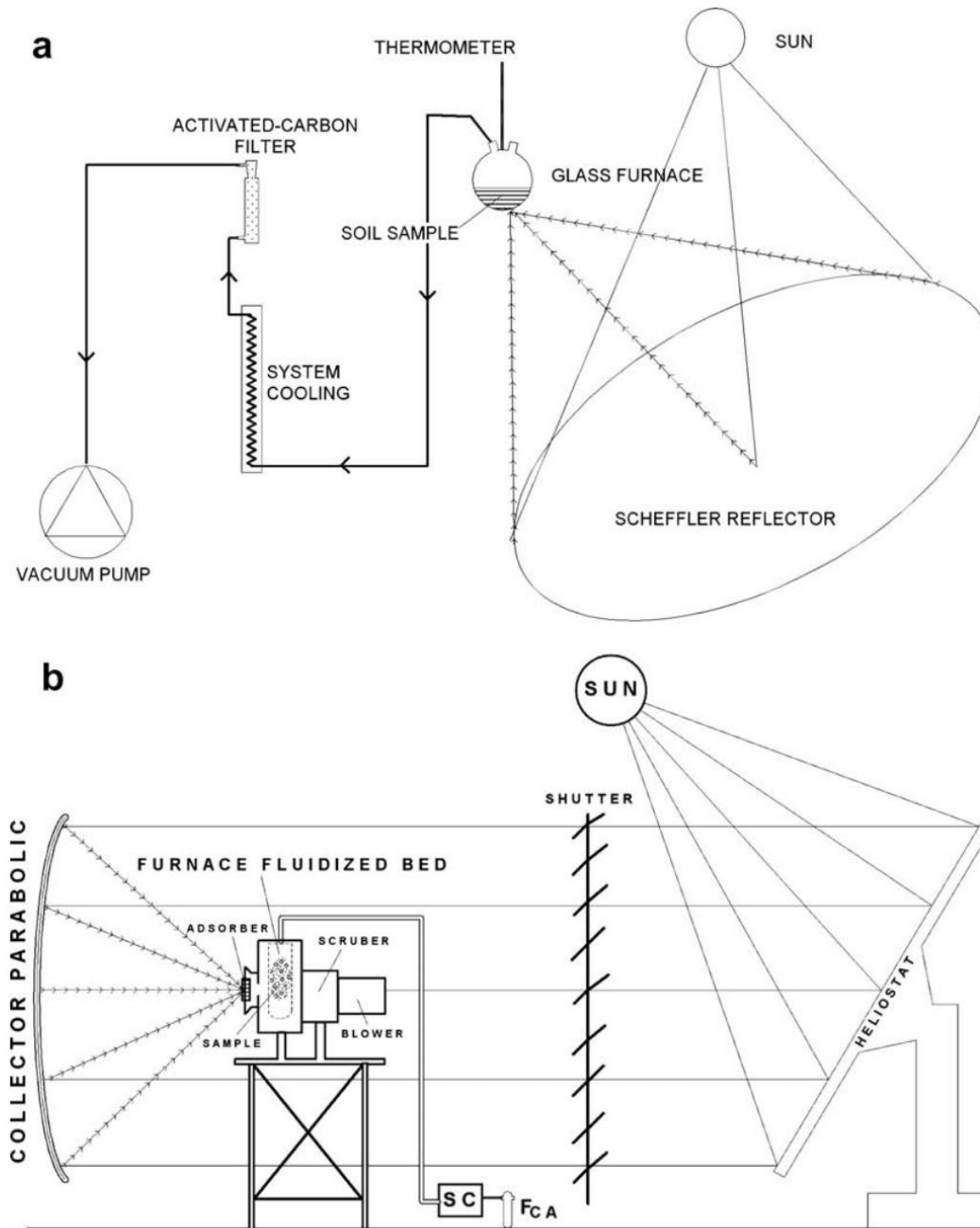


Fig. 1. (a) Scheme of operation of the low-temperature solar furnace (LT-UPC). (b) Scheme of operation of the middle-temperature solar furnace (MT-PSA). SC: system cooling, FCA: activated-carbon filter.

Precipitation

- ▶ Removal of residual mercury from lixiviation process streams
 - ▶ Cyanide dissolves mercury ~30% less effectively than gold
 - ▶ Sulfides (Na_2S , CaS , ZnS), sulfur-based organic substances (Xanthate) highly effective at precipitating mercury
 - ▶ No reduction in gold extraction efficiency
 - ▶ 95-99% removal
 - ▶ Widely available
-



Precipitation

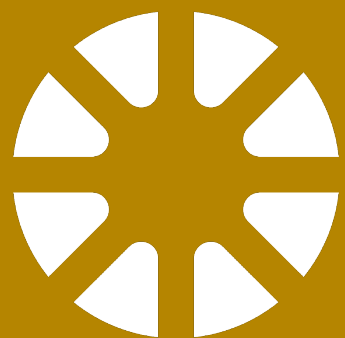
Deployment readiness	Common use in cyanidation processes
Recovery (% range)	95-99%
Tailings compatibility	In CN solutions only
Estimated cost (\$/T)	Cheap, common options exist
Availability	Widely available
Ease of adoption	Normal CN agitated tank facilities should suffice
Health & env risk	Normal industrial chemicals risk
Gender dimensions	Could be applied in micro-cyanidation
Evidence confidence	Widespread industrial use and research
Gov + implementation capacity	Common use in cyanidation processes

Activated carbon

- ▶ Oldest, simplest and most effective way to remove mercury vapour, droplets and ions from air or water
 - ▶ Filters industrial flue gas streams and wastewaters
 - ▶ Made by burning wood in an oxygen poor environment
 - ▶ Molecular scale cracks and crevices whose electrostatic and size properties attract, trap and hold metals
 - ▶ Sulfur-impregnated carbon can typically remove 99.99% of residual mercury from air
 - ▶ Granular activated carbon is used to capture gold and mercury from cyanide solutions
-

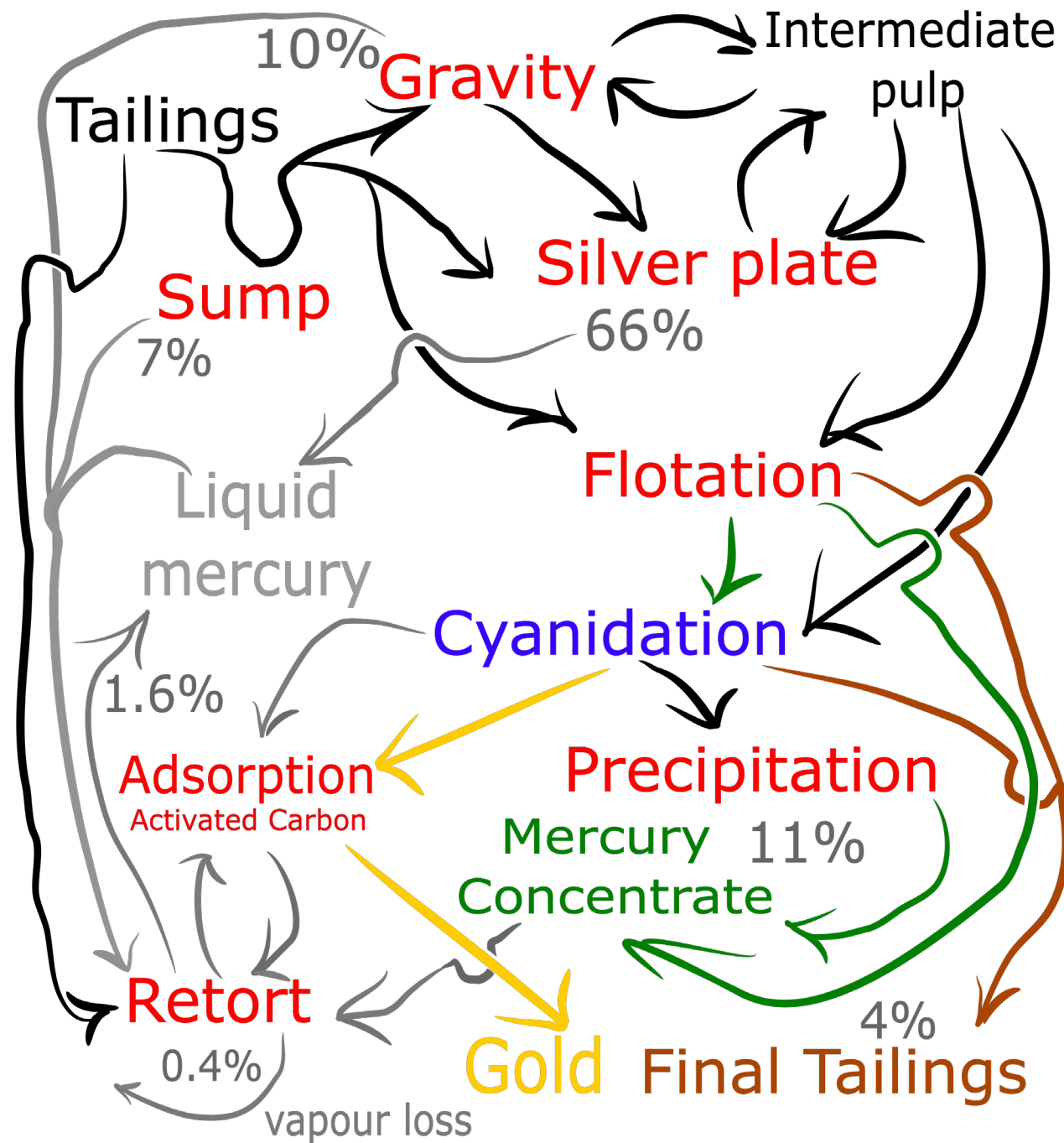
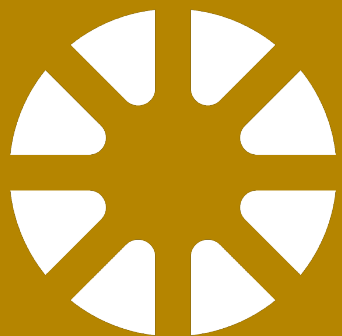
Adsorption with Activated Carbon

Deployment readiness	Common artisanal and industrial use
Recovery (% range)	95-99%
Tailings compatibility	Universal
Estimated cost (\$/T)	Cheapest and simplest
Availability	Made easily by burning wood, coconut shells, etc
Ease of adoption	Widely used for air filtration and in cyanidation
Health & env risk	Safe and non-toxic
Gender dimensions	Accessible
Evidence confidence	Widespread practical use
Gov + implementation capacity	Widespread use in artisanal and industrial processes



Sump





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